SOLVENT FREE OXALIC ACID CATALYZED SYNTHESIS OF 1,5-BENZODIAZEPINES

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ABSTRACT

In the present study 1, 5-benzodiazepines were synthesized from a range of α , β -unsaturated ketones and *o*-phenylendiamine using oxalic acid 10 mol% as a catalyst under solvent free conditions. The yields of the present method are better than the reported method which explains effectiveness of oxalic acid catalyst. The cost effective, resourceful, undemanding and environment friendly are the advantageous aspects of this method.

Keywords: 1, 5-benzodiazepine, oxalic acid, solvent free conditions, α , β -unsaturated ketones, o-phenylenediamine.

1. INTRODUCTION

Benzodiazepines are important class of heterocyclic scaffolds because of their commercial importance and clinical success. The pharmacological activities of benzodiazepines¹ like anxiolytics, hypnotics, tranquilizers, and anticonvulsants, have led to their recognition by medicinal community as the structure of particular significance.² Although the first benzodiazepine was introduced as a drug nearly 40 years ago,³ the research in this area is still very active in view of producing the compounds of enhanced pharmacological activity. In the last decade, the area of biological interest of 1, 5-benzodiazepines has been extended to several diseases such as cancer, viral infection, and cardiovascular disorders.⁴ Moreover 1, 5-benzodiazepines are key intermediates for the synthesis of various fused ring systems such as triazolo-, oxadiazolo-, oxazino- or furanobenzodiazepines.⁵ In addition, benzodiazepine derivatives also have commercial importance as dyes for acrylic fibres in photography.⁶

Most of the methods reported in the literature for the synthesis of benzodiazepines either includes (i) condensation of *o*-phenylenediamine with $\alpha_i\beta$ -unsaturated ketones⁷ or (ii) the reaction of *o*-phenylenediamines with various ketones using variety of reagents and catalysts. These includes, (i) BF₃. Et₂O₈ NaBH₄,⁹ polyphosporic acid,¹⁰ ceric ammonium nitrate (CAN),¹¹ and TCT ¹² in the presence of various solvents; (ii) MgO and POCl₃,¹³ Vb(OTf)₃,¹⁴ Hg(OTf)₃,¹⁵ CeCl₃ 7H₂O/NaI supported on silica gel,¹⁶ Sc(OTf)₃,¹⁷ Er(OTf)₃, InBr₃,¹⁸ sulphated zirconia,¹⁹ ZnCl₂²⁰ and SiO₂/ZnCl₂²¹ under solvent free conditions; (iii) Al₂O₃/P₂O₅²² or acetic acid²³ with microwave irradiation; (iv) Amberlyst-15 in ionic liquids;²⁴ (v) 1,3-di-*n*-butylimidazolium bromide without any catalyst.²⁵ Literature survey also reveals the method of Shrinivasan and co-workers, a multicomponent reaction for the synthesis of 3-*H*benzo[b]-1,4-diazepines.²⁶ Recently, Jiang and co-workers reveals [4+2+1] cycloaddition of *o*-phenylenediamine and ethyl propionate to synthesize 3, 4-disubstituted 1,5-benzodiazepines in good yield.²⁷

Oxalic acid is the compound in which two carboxylic groups are joined together directly and hence supposed to be one of the strongest organic acid. The use of oxalic acid as a catalyst was reported for the deprotection of ketal to give the corresponding aldehydes or ketones.²⁸ It was also found to be used for isomerisation of δ^5 -cholasten-3-one to δ^4 -cholesten-3-one²⁹ and dealumination of zeolite.³⁰ Furthermore, we have used oxalic acid as a catalyst in our previous study for the synthesis of dihydropyrimidine 2-(1*H*)-ones,³¹ 2-aryl-1-arylmethyl-1*H*-benzimidazoles and 2,4,5-triaryl-1*H*-imidazoles,³² quinazolin-4(3*H*)-ones³³ and coumarins.³⁴ In order to explore the use of oxalic acid catalyst further, herein we report the synthesis of 1, 5-benzodiazepines from o-phenylene diamine and various α , β -unsaturated ketones.

2. RESULT AND DISCUSSION

Although development of nonhazardous synthetic methodologies for organic reactions is one latest challenge to the organic chemists, only few of the benzodiazepine synthesis methods tried to answer this demand. Nevertheless, the search for new methodology to synthesize benzodiazepine and analogs continues.

In an attempt to identify the most optimal experimental conditions a detailed study was performed on *o*-phenylenediamine (4) and α , β -unsaturated ketones (3a) (chalcone) (Scheme 2). The reaction condition has been optimized with respect to catalyst loading and use of solvents (Table1). Synthesis of target molecule has been tried using different solvents like DMF, 1, 4-dioxane, THF, etc. Different solvents promote the formation of target compounds with low to moderate yields. The reaction has also been carried out in aqueous conditions where the reaction found to be very sluggish and very trace amount of product has been formed. From the result it is observed that use of 10 mol % catalyst is more favourable, giving the product with 93% yield. However, only modest yield of condensation product was detected at room temperature, probably due to low nucleophilicity of the amino group, as described in previous reports.35 But, good results were registered for the formation of 1, 5 benzodiazepine 5a-j almost in quantitative yield ranging from 72-88% when the reaction was carried out at about 80°C temperature. The highest yield of 88% was registered with ACN whilst, only traces of the product formed when the reaction was carried out in water. Furthermore, the study was also carried out in nonpolar solvents like DCM and chloroform (Table 1, entries 6 and 7), but the yields registered were less with added reaction time at 80°C and no product was detected at room temperature.

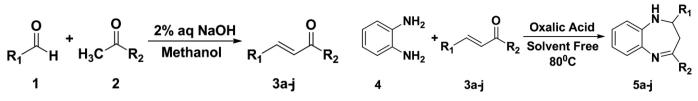
Taking into consideration the green synthetic approach and hazards caused by organic solvents the reaction was further tried under solvent free conditions. Surprisingly, the product formed was registered with 91% yield with 20 mol% of the catalyst and 93% with 10mol% (Table 1, entries 8 and 9 respectively). Further reduction in mol% of the catalyst resulted in lower yields and added reaction time (Table 1, entries 10 and 11).

Therefore, it was concluded that 10 mol% of oxalic acid catalyst was sufficient to form 1, 5-benzodiazepine in almost quantitative yield of the isolated product under solvent free conditions. Hence, the same reaction conditions were followed for the further synthesis of 1, 5-benzodiazepine analogs **5a-j** (Table 2).

In an attempt to study the effect of substituted chalcones in the condensation, we have carried out the condensation of o-phenylenediamine (4) with these substituted chalcones (3a-j) (Scheme 2). The best results for the condensation were registered for the benzodiazepines substituted with electron releasing groups like –OH, -OMe, -CH₃ at *ortho* and *para* positions, however highest yield was registered with benzodiazepine with the –OH group at *ortho* position and –OMe group at *para* position (Table 2, Entry 5j and 5h respectively). In addition to this the reaction was also tried for electron withdrawing substituents which resulted in the the products with moderate yield (Table 2, Entry 5b, 5d and 5f).

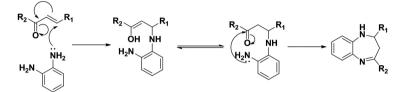
Therefore, we can conclude that we have demonstrated the method to study an efficiency of oxalic acid catalyst by synthesizing an assorted 1,5-benzodiazepines (Table 2, Entry **5a-j**) using **4** and **3a-j** in solvent free conditions. Further, we have also demonstrated observation regarding the effect of different substituents on the quantitative yield of isolated product. Moreover, easy work up, shorter reaction time, good to better yields and environment friendly were the advantages of the anticipated method.

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Scheme 1 Synthesis of α , β -unsaturated ketones **3a-j** from substituted aldehydes and ketones.

Scheme 2 Scheme for the synthesis of 1, 5-benzodiazepines (5a-j) from 3a-j



Scheme 3 Proposed mechanism for the synthesis of benzodiazepine

Table 1 Optimization of reaction conditions and mol% of oxalic acid for the synthesis of 5a (Entry 1, Table 2).

Entry	Solvent	Oxalic acid (mol%)	Reaction time (min)	Yield (%) ^a
1	DMF	20	180	84
2	1,4-dioxane	20	160	72
3	THF	20	150	76
4	ACN	20	155	88
5	H ₂ O	20	180	Trace
6	DCM	20	175	62
7	CHCl ₃	20	185	54
8	Solvent free	20	135	91
9	Solvent free	10	150	93
10	Solvent free	5	185	80
11	Solvent free	2.5	210	77

^a yields are of pure isolated compound

3. EXPERIMENTAL

All reagents and chemicals were of analytical grade and used without further purification.

Melting points of all the compounds were determined using digital melting point apparatus (SRS) and were uncorrected. IR spectrum of all compounds was recorded with FT-IR spectrophotometer (Brucker). ¹HNMR (Varian NMR 400Hz) of the compound **5a-j** was recorded in CDCl₃ using TMS as an internal standard and molecular weight was confirmed by using LC-MS (Scinpor ESI)

3.1 General procedure for the synthesis of 3a-j.

In a round bottom flask, 2.2 g of sodium hydroxide was added in water & methanol. The flask was immersed in a bath of crushed ice & kept on mechanical stirrer. The stirrer was started, freshly distilled acetophenone (0.43 mol) & pure benzaldehyde (0.43 mol) was added in the flask. The mixture was stirred for 4-5 h at 15-20°C. The thick mixture was removed & kept in a refrigerator for overnight. The product was filtered & washed with cold water until the washings were neutral to litmus. Further the product was washed with ice cold rectified spirit, dried & recrystallized from rectified spirit to afford the desired product.

3.2 General procedure for the synthesis of 5a-j

3.2.1 Synthesis of 2,3-dihydro-2,4-diphenyl-1H-benzo[b][1,5]diazepine (5a)

A mixture of α , β -unsaturated ketones **3a** (10mmol), *o*-phenylenediamine **4** (10 mmol) and oxalic acid (10 mol%) was heated at 80°C under solvent free conditions for the appropriate time about 150 min. After completion of reaction (monitored by TLC), the reaction mixture was cooled at room temperature and extracted with diethyl ether (3 × 10ml). The combined organic layers were dried over anhydrous Na₂SO₄. The solvent was removed and the solid residue was purified further on silica gel column (2.5 x 45 cm), pre packed in chloroform. Elution of the column with petroleum ether:ethyl acetate (18:2) gave a homogeneous residue which was concentrated in vacuo to give compound **5a**. Yield: 96%, MP (SRS melting point apparatus): 136-137°C; ¹H NMR (400 MHz, CDCl₃, δ ppm) δ: 1.9 (d, 2H, CH₂), 2.2 (t, 1H, CH), 3.9 (s, 1H, NH), 6.9-7.2 (8H, Ar-H), 7.4-7.8 (4H, Ar-H), 7.8-8 (2H, Ar-H); ¹³C NMR (CDCl₃) δ = 38.2, 63.4, 113.6, 122.4, 125.9, 126.2, 127.3, 127.8, 128.1, 128.9, 129.7, 132.4, 139.9, 142.3, 144.8, 146.4, 163.3; IR (KBr): 1592 (C=C), 1643 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 298.38 (M⁺), 299.25 (M+1); MF: C₃₁H₄N₅.

3.2.2 Compound (5b): 2-(4-chlorophenyl)-2,3-dihydro-4-phenyl-1Hbenzo[b][1,5]diazepine

Yield: 87%, MP: 110-112°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 1.7 (d, 2H, CH₂), 2.9 (t, 1H, CH), 4.2 (s, 1H, NH), 7.18-7.16 (4H, Ar-H), 7.23-7.64 (5H, Ar-H), 6.8-7.2 (2H, Ar-H); ¹³C NMR (CDCl₃) δ = 38.6, 62.1, 115.4, 120.7, 122.2, 125.7, 127.4, 127.9, 128.8, 129.5, 130.6, 133.1, 137.4, 142.2, 143.6, 146.3, 161.8; IR (KBr): 1570 (C=C), 1638 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 332.69 (M⁺), 334.09 (M+2, 33.9%), 333.71 (M+1, 22.8%); MF: C₂₁H₁₇ClN₂.

3.2.3 Compound (5c): Synthesis of 2-(2,3-dihydro-2-(3,4,5-trimethoxyphenyl)-1H-benzo[b] [1,5]diazepin-4-yl)-5-methylphenol

Yield: 89%, MP: 96-98°C; ¹HNMR (400 MHz, CDCl, δ ppm): δ 1.83 (d, 2H, CH₂), 2.73 (t, 1H, CH), 3.85 (s, 1H, NH), 6.18 (2H, Ar-H), 6.65-7.2 (3H, Ar-H), 6.42-7.11 (4H, Ar-H), 4.85 (s, 1H, OH), 2.15 (s, 3H, CH₃), 3.35 (s, 9H, OCH₃); ¹³C NMR (CDCl₄) δ = 20.9, 38.7, 55.3, 62.1, 64.9, 101.2, 113.5, 117.2, 118.8, 122.3, 124.4, 125.7, 126.4, 129.6, 136.6, 139.1, 140.1, 143.4, 146.6, 154.1, 159.9, 164.1; IR (KBr): 3417 (O-H, broad), 1543 (C=C), 1642 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 418.10 (M⁺), 419.45 (M+1, 27.34%); MF: C₂₅H₂₆N₂O₄. 3.2.4 Compound (5d): Synthesis of 2-bromo-6-(2,3-dihydro-2-(3,4,5-

3.2.4 Compound (5d): Synthesis of 2-bromo-6-(2,3-dihydro-2-(3,4,5-trimethoxyphenyl)-1H-benzo[b][1,5]diazepin-4-yl)-4-methylphenol

Yield: 76%, MP: 113-114°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 1.56 (d, 2H, CH₂), 3.12 (t, 1H, CH), 4.01 (s, 1H, NH), 6.22 (2H, Ar-H), 7.07-7.19 (2H, Ar-H), 6.37-7.01 (4H, Ar-H), 4.54 (s, 1H, OH), 2.35 (s, 3H, CH₃), 3.42 (s, 9H, OCH₃); ¹³C NMR (CDCl₃) δ = 19.3, 40.1, 57.3, 62.1, 64.4, 101.3, 113.9, 116.6, 119.7, 123.1, 125.2, 126.9, 128.1, 134.1, 136.2, 137.1, 138.9, 142.6, 146.7, 151.2, 159.3, 163.9; IR (KBr): 3387 (O-H, broad), 1585 (C=C), 1641

(C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 497.10 (M⁺), 498.45 (M+1, 98.34%); MF: C₂, H₂,BrN₂O₄.

3.2.5 Compound (**5e**): Synthesis of 4-(2,3-dihydro-2-phenyl-1H-benzo[b] [1,5]diazepin-4-yl) phenol

Yield: 91%, MP: 122-123°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 2.01 (d, 2H, CH₂), 2.98 (t, 1H, CH), 4.11 (s, 1H, NH), 7.08-7.27 (5H, Ar-H), 6.45-7.35 (4H, Ar-H), 6.3-7.11 (4H, Ar-H), 4.87 (s, 1H, OH); ¹³C NMR (CDCl₃) δ = 20.2, 40.1, 64.3, 113.3, 117.4, 120.2, 124.1, 125.4, 126.9, 127.6, 130.8, 135.2, 138.3, 142.9, 143.8, 146.2, 159.5, 164.3; IR (KBr): 3413 (O-H, broad), 1592 (C=C), 1636 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 314.10 (M⁺); MF: C₂₁H₁₈N₂O.

3.2.6 Compound (**5f**): Synthesis of 2,3-dihydro-2,4-bis(4-nitrophenyl)-1H-benzo[b][1,5] diazepine

Yield: \$2%, MP: $131-132^{\circ}$ C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 1.85 (d, 2H, CH₂), 2.76 (t, 1H, CH), 3.96 (s, 1H, NH), 7.21-8.15 (4H, Ar-H), 7.7-8.35 (4H, Ar-H), 6.32-7.07 (4H, Ar-H); ¹³C NMR (CDCl₃) δ = 37.9, 62.1, 113.2, 120.5, 122.3, 124.4, 125.1, 126.2, 127.4, 129.2, 140.3, 142.7, 144.2, 145.3, 147.9, 149.1, 163.9; IR (KBr): 1351, 1555 (NO₂), 1577 (C=C), 1648 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 388.74 (M⁺), 389.23 (M+1, 23.33%) MF: C₂, H₄N₄O₄.

MF: $C_{21}H_{16}N_4O_4$. 3.2.7 Compound (**5g**): Synthesis of 2,3-dihydro-2,4-dip-tolyl-1H-benzo[b] [1,5]diazepine

Yield: 92%, MP: 130-132°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 2.11 (d, 2H, CH₂), 2.99 (t, 1H, CH), 4.22 (s, 1H, NH), 6.98-7.15 (4H, Ar-H), 7.09-7.54 (4H, Ar-H), 6.4-7.02 (4H, Ar-H), 2.23 (s, 6H, CH₃); ¹³C NMR (CDCl₃) δ = 20.4, 40.3, 61.7, 113.3, 120.2, 124.4, 125.7, 126.5, 128.1, 128.6, 130.3, 135.6, 136.8, 137.3, 139.9, 142.8, 146.3, 163.8; IR (KBr): 1586 (C=C), 1643 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 326.34 (M⁺), 327.23 (M+1, 26.13%)

MF: C₂₃H₂₂N₂.

3.2.8 Compound (5h): Synthesis of 2,3-dihydro-2,4-bis(4-methoxyphenyl)-1H-benzo[b][1,5] diazepine

Yield: 95%, MP: $112-114^{\circ}$ C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 2.06 (d, 2H, CH₂), 3.11 (t, 1H, CH), 4.05 (s, 1H, NH), 6.66-7.22 (4H, Ar-H), 6.76-7.62 (4H, Ar-H), 6.4-7.02 (4H, Ar-H), 3.67 (s, 6H, OCH₃); ¹³C NMR (CDCl₃) δ = 40.1, 54.7, 63.7, 112.9, 113.7, 115.9, 124.1, 125.2, 126.1, 127.1, 129.9, 131.7, 140.3, 143.1, 146.7, 157.3, 161.2, 164.5; IR (KBr): 1587 (C=C), 1639 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/e): m/e 358.43 (M⁺), 359.33 (M+1, 25.13%); MF: C₂₃H₂, N₂O₂.

3.2.9 Compound (5i): Synthesis of 2-(2,3-dihydro-2-phenyl-1H-benzo[b] [1,5]diazepin-4-yl) phenol

Yield: 84%, MP: 128-129°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 1.84 (d, 2H, CH₂), 3.72 (t, 1H, CH), 4.15 (s, 1H, NH), 7.02-7.22 (5H, Ar-H), 6.66-7.40 (4H, Ar-H), 6.3-7.02 (4H, Ar-H), 4.89 (s, 1H, OH); ¹³C NMR (CDCl₃) δ = 40.6, 64.7, 113.2, 116.5, 117.6, 119.9, 123.4, 124.9, 125.5, 126.1, 127.2, 129.8, 130.7, 133.8, 139.2, 142.8, 146.7, 161.3, 164.2; IR (KBr): 3410 (O-H, broad), 1590 (C=C), 1640 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 314.65 (M⁺); MF: C₂₁H₄N₂O.

3.2.10 Compound (5j): Synthesis of 2-(2,3-dihydro-2-(4-methoxyphenyl)-1H-benzo[b][1,5] diazepin-4-yl)phenol

Yield: 96%, MP: 107-108°C; ¹HNMR (400 MHz, CDCl₃, δ ppm): δ 1.74 (d, 2H, CH₂), 3.83 (t, 1H, CH), 4.03 (s, 1H, NH), 6.67-7.0 (4H, Ar-H), 6.7-7.18 (4H, Ar-H), 6.38-7.0 (4H, Ar-H), 5.1 (s, 1H, OH), 3.7 (s, 3H, OCH₃); ¹³C NMR (CDCl₃) δ = 40.7, 56.5, 64.2, 112.9, 115.7, 116.5, 117.6, 120.2, 121.1, 124.9, 125.7, 126.6, 130.8, 131.6, 132.1, 142.7, 144.9, 157.3, 161.5, 163.9; IR (KBr): 3415 (O-H, broad),1591 (C=C), 1639 (C=N) cm⁻¹; LC-MS (Scinpor ESI m/z): 344.23 (M⁺), 345.12 (M+1, 24.1%); MF: C₂,H₂₀N₂O,

Entry	R ₂	R ₁	Reaction time (min)	Yield (%) ^a	Melting Point(⁰ C)
5a			150	93	136-137
5b		CI	160	87	110-112
5c	HO CH3	H ₃ CO H ₃ CO OCH ₃	155	89	96-98
5d	HO Br CH ₃	H ₃ CO H ₃ CO OCH ₃	160	76	113-114
5e	ОН		150	91	122-123
5f	NO ₂	O ₂ N	140	82	131-132
5g	CH3	H ₃ C	135	92	130-132
5h	OCH3	H ₃ CO	130	95	112-114
5i	но		155	84	128-129
5j	но	H ₃ CO	165	96	107-108

 Table 2 Oxalic acid catalyzed synthesis of 5a-j.

^a Yield are of pure isolated compounds.

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