

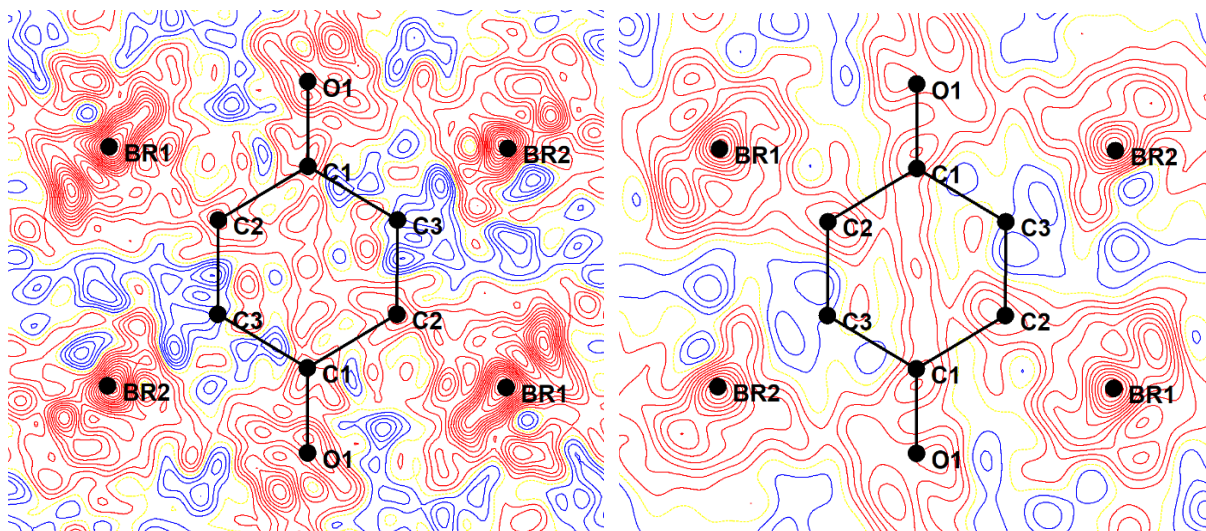
# IUCrJ

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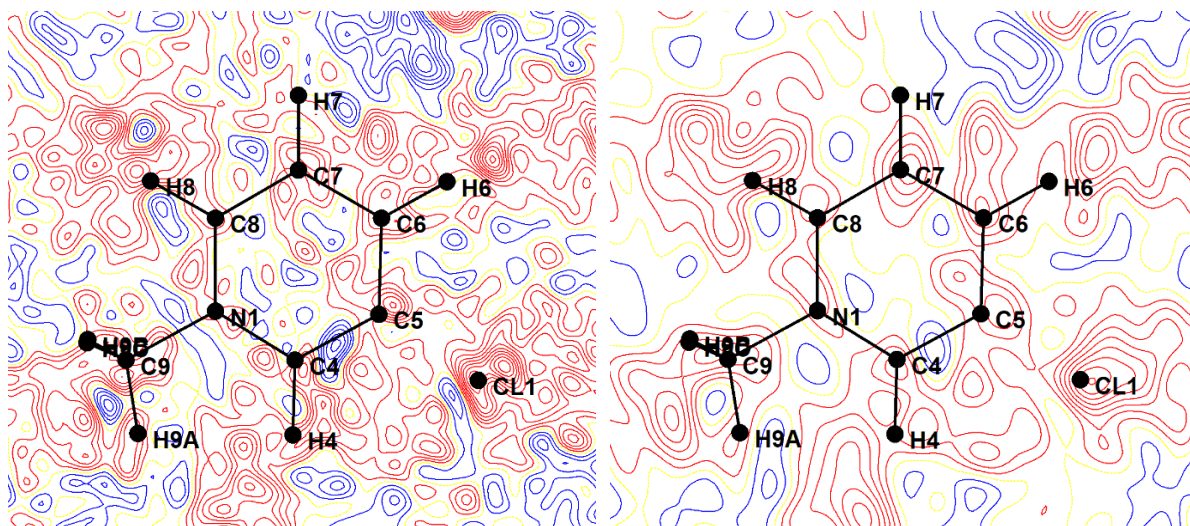
**Supporting information for article:**

**The nature of  $\pi$ -hole interaction between iodide anion and quinoid ring in the crystalline state**

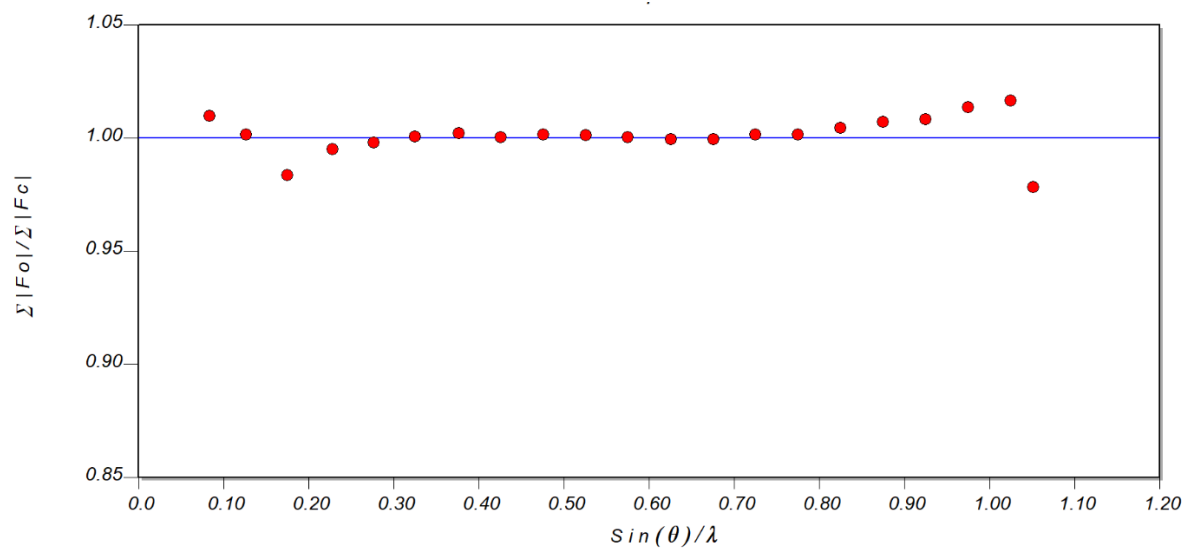
**Valentina Milašinović, Vedran Vuković, Anna Krawczuk, Krešimir Molčanov, Christoph Hennig and Michael Bodensteiner**

**S1. Residual densities and other details on refinement**

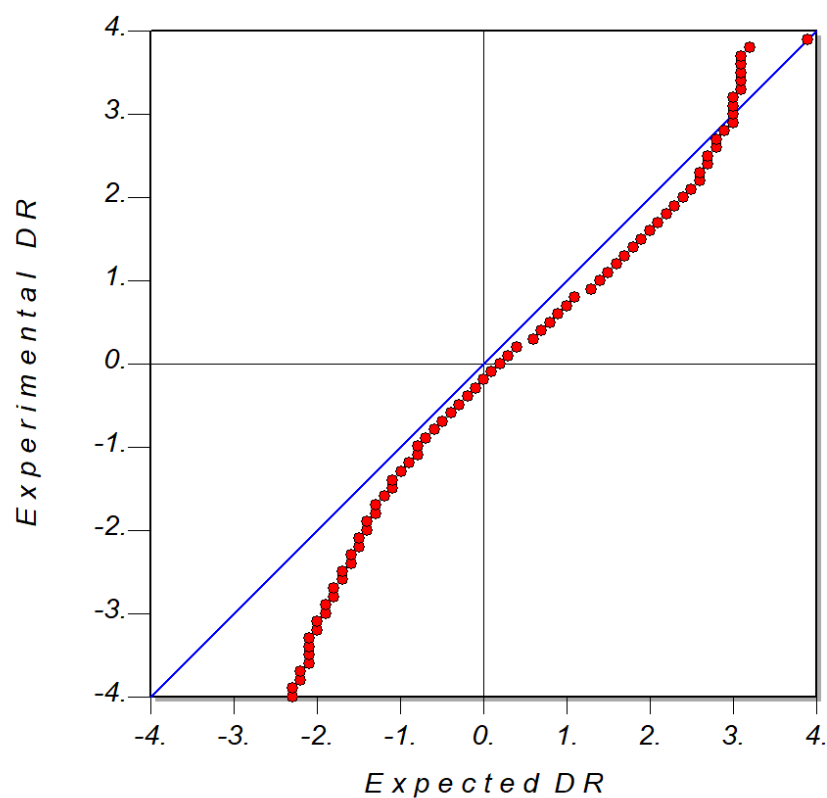
**Figure S1** Residual density in the mean plane of Br<sub>4</sub>Q with a) all reflections used and b) only low-angle reflections ( $\sin \theta / \lambda < 0.7 \text{ \AA}^{-1}$ ) used. Positive density is shown in blue and negative in red; yellow dotted lines represent zero density. The spacing between contours is of  $0.05 \text{ e\AA}^{-3}$ .



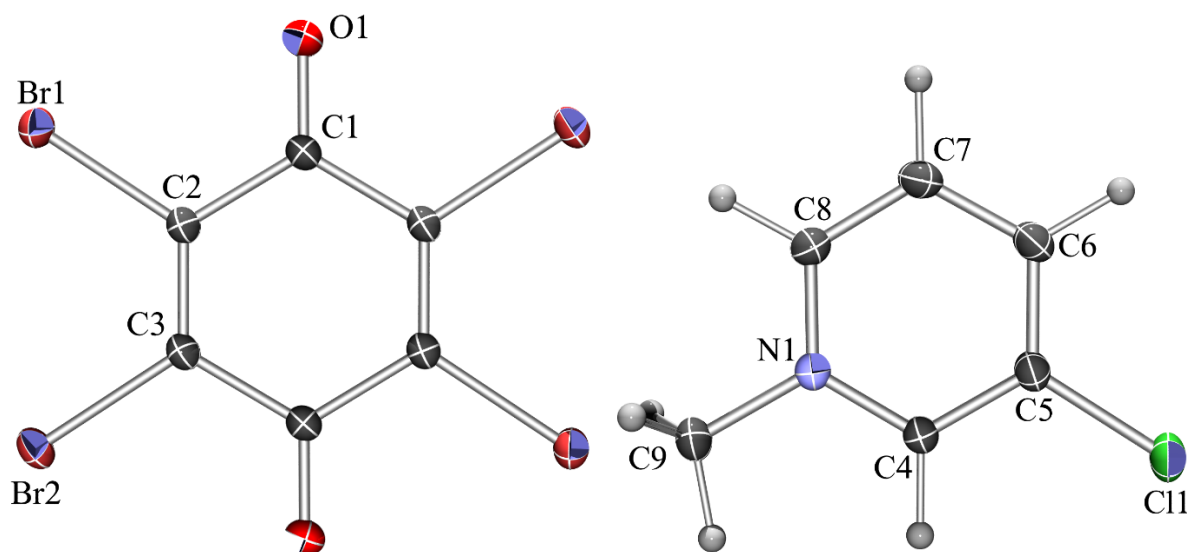
**Figure S2** Residual density in the mean plane of 3-Cl-N-MePy cation with a) all reflections used and b) only low-angle reflections ( $s < 0.7 \text{ \AA}^{-1}$ ) used. Positive density is shown in blue and negative in red; yellow dotted lines represent zero density. The spacing between contours is of  $0.05 \text{ e\AA}^{-3}$ .



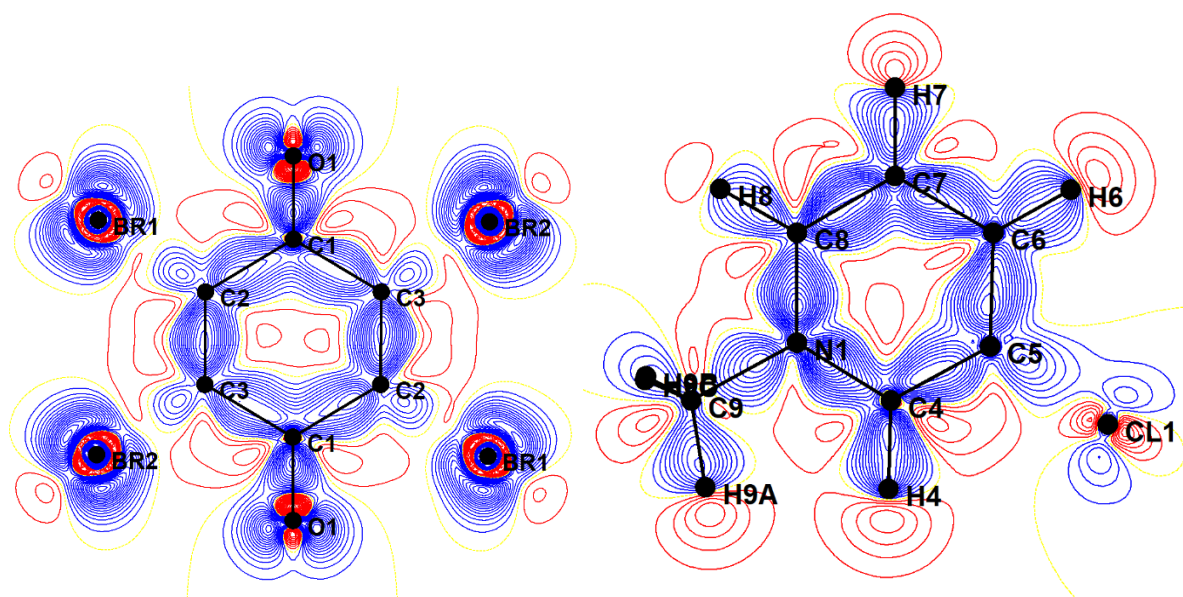
**Figure S3** XDRK plot showing the fit of  $\langle Y_{\text{obs}} \rangle$  vs.  $\langle Y_{\text{calc}} \rangle$  as a function of resolution.



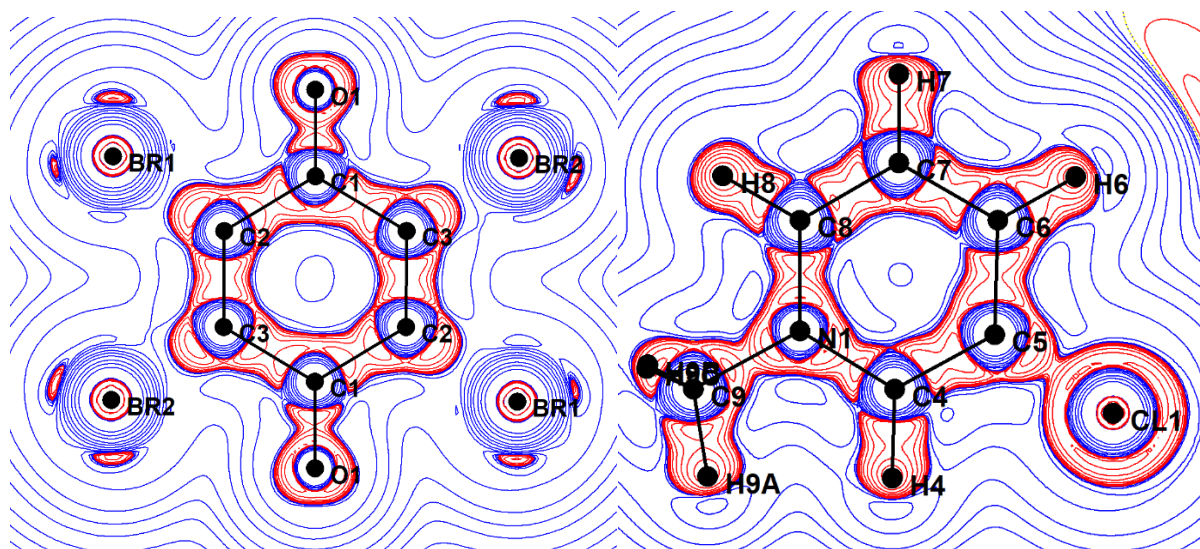
**Figure S4** XDRK plot showing the expected and experimental  $Y_{\text{obs}} - Y_{\text{calc}}$  data profile.

**S2. ORTEPs, deformation densities, Laplacians and electrostatic potential**

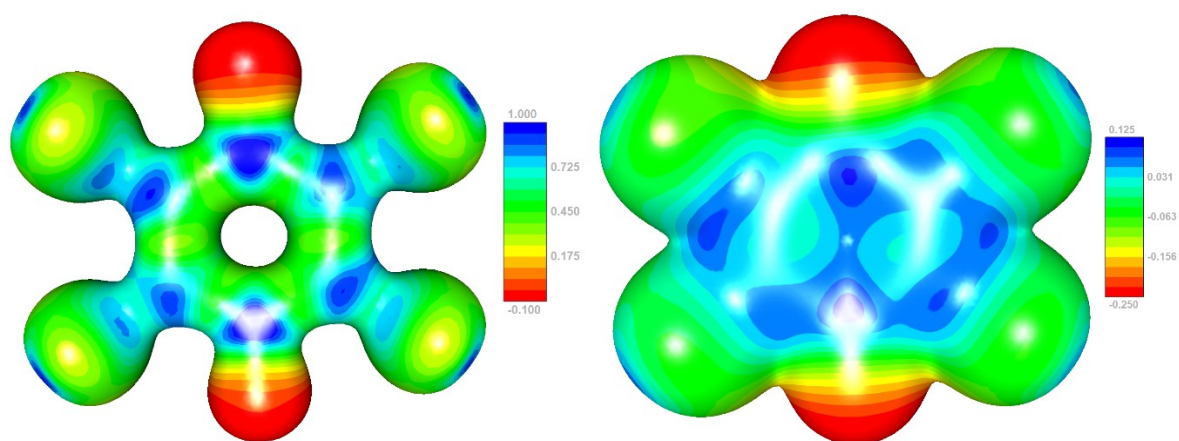
**Figure S5** ORTEP-3 drawings of a) Br<sub>4</sub>Q and b) 3-Cl-N-MePy cation. Displacement ellipsoids are drawn for the probability of 50% and hydrogen atoms are shown as spheres of arbitrary radii.



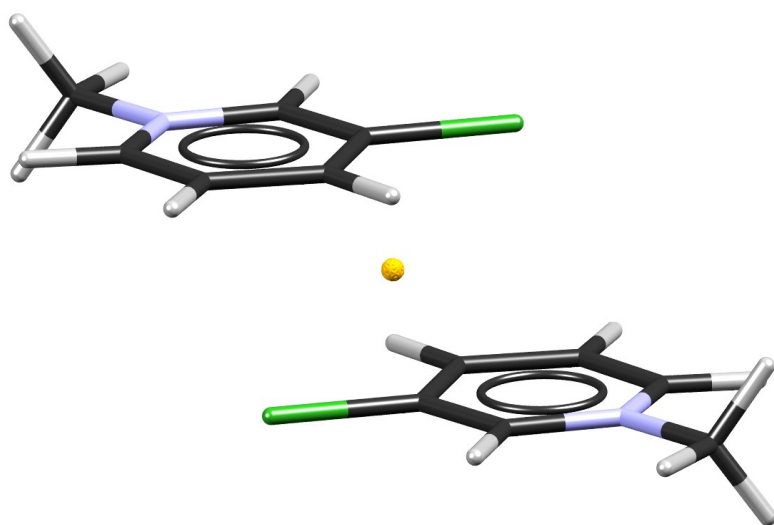
**Figure S6** Deformation density maps in the mean planes of a) Br<sub>4</sub>Q and b) 3-Cl-N-MePy cation. The spacing between contours is of 0.05 e Å<sup>-3</sup>; the positive density is blue, the negative is red and the zero contour is drawn as a yellow dotted line.



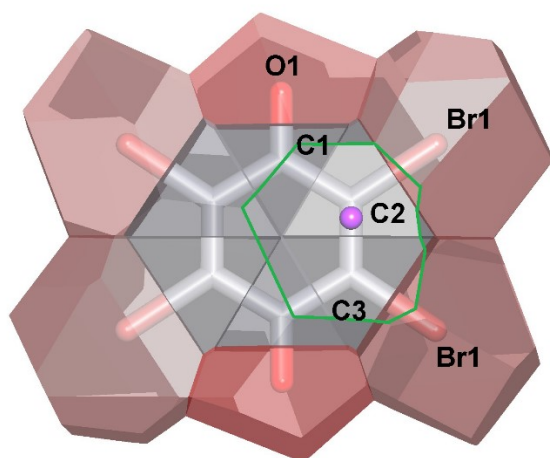
**Figure S7** Laplacian of electron density plotted in the mean planes of a) Br<sub>4</sub>Q and b) 3-Cl-N-MePy cation. The contours are drawn for  $2, 4, 8 \cdot 10^n \text{ e } \text{Å}^{-5}$ ,  $n = -3 \dots 2$ ; positive Laplacian is blue and negative is red.



**Figure S8** Electrostatic potential of Br<sub>4</sub>Q molecule plotted onto an electron density isosurface of a)  $0.5 \text{ e } \text{Å}^{-3}$  and b)  $0.05 \text{ e } \text{Å}^{-3}$ .

**S3. Other geometric data**

**Figure S9** A pair of antiparallel stacked 3-chloro-*N*-methylpyridinium cations related by an inversion centre (shown as a yellow sphere).



**Figure S10** Surface of a Br<sub>4</sub>Q molecule comprised of VDPs. Area corresponding to the contact with the iodide (shown as a purple sphere near C2) is highlighted in green.

**S4. Topology of electron density****Table S1** Topology of electron density in Br<sub>4</sub>Q, derived from electron-density after multipole refinement and periodic calculations (in italic).

Bond	Length (Å)	Electron Density	Laplacian	Ellipticity	Bond order
		(eÅ <sup>-3</sup> ) $\rho_{cp}$	(eÅ <sup>-3</sup> )		$n_{topo}$
C1–O1	1.2130(13)	2.874	–29.4	0.17	1.50
		<i>2.675</i>	<i>–2.410</i>		
C1–C2	1.4909(10)	1.790	–12.3	0.10	0.86
		<i>1.769</i>	<i>–14.459</i>		
C2–C3	1.3462(9)	2.307	–22.5	0.27	1.65
		<i>2.201</i>	<i>–19.038</i>		
C3–C1	1.4898(10)	1.840	–13.1	0.09	0.90
		<i>1.773</i>	<i>–14.459</i>		
C2–Br1	1.8778(7)	1.001	1.7	0.26	
		<i>1.131</i>	<i>–4.097</i>		
C3–Br2	1.8767(6)	0.990	1.7	0.13	
		<i>1.134</i>	<i>–4.097</i>		

**Table S2** Topology of electron density in 3-Cl-N-MePy cation, derived from electron-density after multipole refinement and periodic calculations (in italic).

Bond	Length (Å)	Electron Density	Laplacian	Ellipticity	Bond order
		(eÅ <sup>-3</sup> ) $\rho_{cp}$	(eÅ <sup>-3</sup> )		$n_{topo}$
N1–C4	1.3497(9)	2.256	–21.7	0.16	1.30
		<i>2.124</i>	<i>–15.905</i>		
N1–C8	1.3469(10)	2.356	–26.0	0.21	1.29
		<i>2.132</i>	<i>–15.423</i>		
N1–C9	1.4745(11)	1.749	–10.6	0.04	1.04

		<i>1.603</i>	<i>-10.603</i>		
C4-C5	1.3833(10)	2.049	-18.5	0.15	1.29
		<i>2.103</i>	<i>-18.315</i>		
C5-C6	1.3924(11)	2.121	-17.0	0.18	1.47
		<i>2.054</i>	<i>-17.592</i>		
C6-C7	1.3889(12)	2.119	-19.9	0.19	1.29
		<i>2.041</i>	<i>-17.110</i>		
C7-C8	1.3865(12)	2.272	-21.1	0.17	1.50
		<i>2.083</i>	<i>-18.074</i>		
C5-C11	1.7262(7)	1.401	-1.3	0.08	
		<i>1.346</i>	<i>-6.266</i>		
C4-H4	1.08	1.820	-19.2	0.09	0.90
		<i>1.892</i>	<i>-22.412</i>		
C6-H6	1.08	1.701	-16.9	0.03	0.99
		<i>1.869</i>	<i>-21.201</i>		
C7-H7	1.08	1.681	-19.8	0.09	0.80
		<i>1.864</i>	<i>-21.063</i>		
C8-H8	1.08	1.550	-13.6	0.12	0.82
		<i>1.890</i>	<i>-21.914</i>		
C9-H9A	1.08(15)	1.725	-16.8	0.14	0.96
		<i>1.892</i>	<i>-20.966</i>		
C9-H9B	1.08(11)	1.443	-7.8	0.16	0.92
		<i>1.879</i>	<i>-20.484</i>		
C9-H9C	1.08(19)	1.692	-16.8	0.15	0.93
		<i>1.885</i>	<i>-20.868</i>		

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