

- 1 "Salt reduction in slow fermented sausages affects the generation of aroma active
- 2 compounds"
- 3 Sara Corral, Ana Salvador, Mónica Flores*
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- 5 Instituto de Agroquímica y Tecnología de Alimentos (IATA-CSIC) Avda. Agustín Escardino 7,
- 6 46980 Paterna, Valencia, Spain
- 7
- 8 Corresponding author. Tel.: +34 96 3900022; fax: +34 96 3636301
- 9 E-mail address: mflores@iata.csic.es (M. Flores).
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12 Abstract

Slow fermented sausages with different salt content were manufactured: control (2.7 % NaCl, S), 16 % salt reduced (2.26 % NaCl, RS) and 16 % replaced by KCl (2.26 % NaCl and 0.43 % KCI, RSK). The effect of salt reduction on microbiology and chemical parameters, sensory characteristics, texture and volatile compounds was studied. The aroma compounds were identified by GC-MS and olfactometry analyses. Small salt reduction (16%) (RS) affected sausage quality producing a reduction in the acceptance of aroma, taste, juiciness and overall quality. The substitution by KCI (RSK) produced the same acceptability by consumers as for high salt (S) treatment except for the aroma that was not improved by KCI addition. The aroma was affected due to the reduction in sulfur and acids and the increase of aldehyde compounds. Aroma compounds that characterized the high salt treatment (S) were dimethyl trisulfide, 3-methyl thiophene, 2,3-butanedione, 2-nonanone and acetic acid. Keywords: fermented sausages, salt reduction, volatile compounds, aroma, flavour.

43 **1. Introduction**

The relation between high salt intake and incidence/prevalence of hypertension has led the European Union (EU) to implement salt reduction initiatives in the EU framework (European Commission, 2008). EU proposed salt reduction of 16 % in 4 years, decreasing 4 % per year in order to allow consumers to adapt to the slightly decreasing salty taste. In some products, salt reduction means lower salty taste, but others products, such as dry curing and processed meats, can lead to safety and technological problem.

50 Salt is an essential ingredient in dry fermented sausages; it is involved in myofibrillar protein solubilization, improve texture; decrease water activity (a_w) controlling the growth of 51 52 pathogens microorganism and finally, it controls the biochemical and enzymatic reactions during ripening, affecting the final flavour (Ruusunen & Puolanne, 2005). The reduction of salt 53 in fermented meat products has been studied through different strategies such as the use of 54 KCl alone (Gou, Guerrero, Gelabert, & Arnau, 1996) or together with other chloride salts 55 56 (CaCl₂, MgCl₂) (Gimeno, Astiasarán & Bello, 1998; Zanardi, Ghidini, Conter & Ianieri, 2010) and also different flavour enhancers have been used (lactate, amino acids and yeast extracts) (Gou 57 58 et al., 1996; Guàrdia, Guerrero, Gelabert, Gou & Arnau, 2008; Campagnol, dos Santos, 59 Wagner, Terra & Pollonio, 2011). However, KCI provides metallic or bitter tastes when it is used at concentrations equal or higher than 40% (Gou et al., 1996, Gelabert, Gou, Guerrero & 60 Arnau, 2003). Furthermore, sausage texture is affected depending on the type of salts used in 61 the substitution. When KCI is used alone it produced an increase in sausage hardness (Gou et 62 al., 1996; Guardia et al., 2008) while the use of KCI in combination with other divalent salts 63 resulted in a decrease in hardness (Gimeno, Astiasarán & Bello, 1999). 64

Moreover, other sensory characteristics are affected by salt substitution such as a 65 decrease in aroma and taste when KCI is used at high percentages (>40%) (Guardia et al., 66 2008, Campagnol et al., 2011). Nevertheless, the effect of salt substitution on aroma has been 67 poorly studied only Campagnol et al. (2011) studied the volatile compounds generated in 68 fermented sausages when NaCl was substitute by KCl and yeast extracts. These authors 69 reported few differences in aroma among sausages when NaCl was reduced in a 25% but after 70 71 50% substitution, the decrease in aroma and taste was evident. In addition, other studies performed on sausage models indicated that salt modifications affected volatile compounds but 72 it depended on ripening time and the type of starter culture used (Olesen, Meyer & Stahnke, 73 2004; Tjener, Stahnke, Andersen & Martinussen, 2004). In contrast, Ravyts, Steen, Goemaere, 74 75 Paelinck, De Vuyst & Leroy (2010) indicated that modifications of salt concentrations in sausages produced a very limited impact on the growth and composition of the microbiota 76 without detecting an effect on volatile composition. However, all these previous studies did not 77 78 evaluate the effect of salt reduction on aroma active compounds as they mainly focused on 79 several volatile compounds.

80 Furthermore, these studies were done in fermented sausages but there are no reports 81 about the reduction of NaCl content in slow fermented sausages (i.e. Chorizo de Cantimpalos, Cacciatore salami, Hungarian type salami, and others). These slow fermented sausages are 82 typically produced in Southern European countries by using low temperatures during ripening 83 (Flores, 1997) and the rate of acidification is low allowing the activity of acid-sensitive bacteria 84 (micrococcaceae and staphylococci). The flavour of these sausages is mostly formed by 85 endogenous or bacterial enzymatic activities and the oxidation of the lipid fraction (Ravyst et al., 86 87 2010). Nothing is known about the effect of salt reduction on volatile aroma compounds in slow 88 fermented sausages. Therefore, it is necessary to evaluate if salt reduction may affect flavour 89 quality because the acceptability of slow fermented sausages is largely dependent upon its flavour (Flores, 1997). NaCl plays an important role in flavour development, since it provides 90 91 the salty taste, enhances savory and meaty flavours and improves the release of volatile aroma 92 compound from the food matrix (Ruusunen & Puolanne, 2005). However, the adaptation to less 93 salty taste by consumers is important as it can be a way to reduce salt content in meat products. For all these reasons, the aim of this work was to study the effect of a 16 % salt 94 95 reduction in the production of aroma active compounds in slow fermented sausages and to 96 determine the effects produced by KCI used during salt reduction.

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98 **2. Materials and methods**

99 **2.1 Dry fermented sausages preparation**

100 Three treatments of dry fermented sausages were manufactured with different salt 101 contents; control treatment (S) with 2.7 % NaCl, low salt treatment (RS) with 2.26 % NaCl and a 102 third treatment (RSK) with 2.26 % NaCl and 0.43 % KCl.

103 Sausages was prepared with lean pork (75 %) and pork back fat (25 %) and the following additives (g/Kg): lactose (30); dextrin (10); sodium caseinate (20); glucose (7); sodium 104 ascorbate (0.5); sodium nitrite (0.15); potassium nitrate (0.15) and starter culture (0.1) SP318 105 TEXEL SA-301 (Danisco, Cultor, Madrid, Spain) containing Lactobacillus sakei, Pediococcus 106 pentosaceus, Staphylococcus xylosus and Staphylococcus carnosus. The manufacturing 107 108 process was the same as described (Olivares, Navarro & Flores, 2010). The meat mixture was stuffed into collagen casings of 9.5 cm diameter (FIBRAN, S.A., Girona, Spain) and the 109 sausages were subjected to drying at 10-12 °C and 70-85 % HR for 57 days. In order to control 110 the ripening process, weight losses and pH were measured during processing (Olivares et al., 111 112 2010).

From each treatment, a 200 g portion of the meat mixture (0 days) and three sausages at 9, 29 and 57 days were randomly collected to study the effect of ripening time and formulation. A 150 g portion of the sample was minced and used for moisture, water activity and pH tests. In addition, sausage color was measured and a 10 g portion was taken for microbiological analysis. The remaining minced sample was vacuum packed and frozen at -20 ¹¹⁸ ^oC for subsequent analyses (TBARS, lipid, protein and ions content). At 57 days, several slices ¹¹⁹ (1 cm thickness) were wrapped in aluminum foil, vacuum packaged and stored at -80 ^oC for ¹²⁰ volatile and aroma analyses. All results were expressed as the mean of three replicates at each ¹²¹ sampling time. Finally, the texture and sensory analysis were carried out at the end of the ¹²² drying process (57 days).

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124 **2.2 Chemical analysis**

pH was measured by introducing a pH meter HI 99163 (Hanna Instruments Inc.,
Hoonsocket, USA) into a mixture of sausage and distilled water (1:1) (ISO 2917:1999). Water
activity was determined using a Fast-lab water activity meter (Gbx, Romans sur Isère Cédex,
France) as described Olivares et al., (2010). Color evaluation was made through the CIE L*, a*,
b* space. The color of the sausages was measured using a colorimeter CR-400/410 (Konica
Minolta Sensing Inc., Japan) with D65 illuminant (Olivares et al., 2010).

Moisture content was determined after dehydration at 100 °C to a constant weight, according to the official method of analysis of meat products (BOE, 1979). Total lipids were extracted from 5 g of minced sausage according to the method of Folch, Lees & Sloane Stanley (1957), using dichloromethane: methanol (2:1) instead of chloroform: methanol (2:1) as solvent. The extract obtained was evaporated in a rotating vacuum evaporator and weighed to determine the total lipid content. Nitrogen content was determined by the Kjeldahl method and protein was estimated by multiplying the nitrogen content by a factor of 6.25.

Thiobarbituric acid reactive substances (TBARS) were quantified to determine the degree of lipid oxidation, as described Olivares, Navarro & Flores (2011) using trichloroacetic acid as solvent instead of perchloric acid. The results were expressed as mg malonaldehyde (MDA)/ kg in dry matter.

Cations (sodium, potassium) were analyzed by ion chromatography as described 142 Armenteros (2009). Chloride anion in sample solutions was determined by using Metrohm 761 143 Compact IC with Metrohm 833 IC Liquid Handling Suppressor unit to improve chromatographic 144 signal. Guard column A-Supp 4/5 (5.0 x 4.0 mm) and analytical column Supp 5-250 (4.0 x 250 145 146 mm) were used to analyze chloride anion. The mobile phase consisted of 1 mM NaHCO₃ and 3.2 mM Na₂CO₃ with 30 ml/l acetone. The concentration of each ion was determined from 147 respective calibration curves, using a set of standard solutions of Na⁺, K⁺ and Cl⁻ (Fluka, 148 Switzerland, Sigma, St. Louis, MO). The results (means of three determinations) were 149 150 expressed as mg/100 g of sample in dry matter.

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152 **2.3 Microbiological analysis**

153 Minced sausage sample was aseptically homogenized with peptone water (1/10) in a 154 Stomacher (IUL Instruments, Barcelona, Spain) for 1 min and decimal dilutions were prepared. 155 Lactic acid bacteria population was determined by the overlay technique to promote anaerobic growth using MRS agar (Scharlau Chemie SA, Barcelona, Spain). *Staphylococci* counts were
obtained on Mannitol salt agar (Scharlau Chemie SA, Barcelona, Spain). Both mediums were
incubated at 30 °C for 3 days.

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160 **2.4 Analysis of volatile compounds**

161 **2.4.1 Gas chromatography-mass spectrometry (GC-MS)**

An Agilent HP 7890 series II GC (Hewlett- Packard, Palo Alto, CA) with an HP 5975C mass selective detector (Hewlett-Packard) equipped with Gerstel MPS2 multipurpose sampler (Gerstel, Germany) was used in all experiments. Extraction of headspace volatile compounds was performed using a solid-phase microextraction (SPME) with an 85 µm Carboxen/ Polydimethylsiloxane (CAR/PDMS) fibre for automatic holder (Supelco, Bellefonte, PA). Before the analysis, the fibre was preconditioned as indicated by the manufacturer.

168 For each experiment, 5 g of dry fermented sausages was minced and weighted into a 169 20 ml headspace vial sealed with a PTFE faced silicone septum and 0.75 mg of BHT was added. The vial was maintained at 37 °C during 30 min to equilibrate its headspace. Then, the 170 171 SPME fibre was exposed to the headspace while maintaining the sample at 37 °C during 3 h. 172 Before each injection, the fiber was baked at 250 °C for 15 min. The compounds adsorbed by 173 the fibre were desorbed in the injection port of the GC-MS for 5 min at 240 °C with purge valve 174 off (splitless mode). The analysis of volatile compounds in the GC-MS was done as described 175 Olivares et al. (2011). The compounds were identified by comparison with mass spectra from 176 the library database (Nist'05), kovats retention index (Kovats, 1965) and by comparison with authentic standards. The quantification of volatile compounds was done in SCAN mode using 177 178 either total or extracted ion chromatogram (TIC or EIC) on an arbitrary scale.

179 **2.4.2 Gas-chromatrography-olfactometry**

A gas chromatograph (Agilent 6890, USA) equipped with a FID detector and sniffing 180 port (ODP3, Gerstel, Mülheim an der Ruhr, Germany) was used to analyze aroma compounds 181 as described Olivares et al. (2011) using SPME technique. The detection frequency method 182 was used to estimate the aromatic impact of each volatile (Pollien, Ott, Montigon, Baumgartner, 183 184 Muñoz-Box & Chaintreau, 1957). Each assessment was carried out according to Olivares et al. 185 (2011). Four trained panellists evaluated the odours from the GC-effluent. Each assessor evaluated 3 sausages per treatment (57 d of ripening), therefore a total of 12 assessments 186 were carried out. The final detection frequency value (DF) for each compound was obtained by 187 188 summation of the 12 sniffings. The detection of an odor by less than three assessors was 189 considered to be noise.

190 Compounds were identified using the following techniques: comparison with mass 191 spectra, comparison with kovats retention indices of authentic standards injected in the GC-MS 192 and GC-O, and by coincidence of the assessors's descriptors with those in the Fenaroli's 193 handbook of flavour ingredients (Burdock, 2002). 194

195 **2.5 Texture profile analysis**

Texture profile analysis (TPA) was performed using TA-XT.plus Texture Analyzer with Texture Exponent software (version 2.0.7.0. Stable Microsystems, Godalming, UK). Four dry fermented sausage slices (diameter 3.5 cm and thick 1.5 cm) of three sausages per treatment were compressed twice to 50 % of their original height as described Olivares et al. (2010). TPA curves were obtained and the main parameters of texture were calculated: hardness, springiness, cohesiveness and as secondary parameter chewiness.

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203 2.6 Sensory analysis

Testing was carried out in a sensory laboratory equipped with individual booths (ISO 204 205 8589, 1988). A panel of 85 untrained consumers was used. The casing was removed and the 206 sausages were cut into slices of 4 mm thickness. Samples from each treatment (S, RS, RSK) 207 were labeled with random, three-digit codes and presented on a plate at room temperature with 208 water and bread without salt to cleanse the palate between samples. An acceptability test was 209 carried out using 9-box hedonic scale (1extremely dislike - 9 extremely like). The attributes 210 evaluated were: appearance, flavour, taste, hardness, juiciness and overall quality. One slice of 211 each treatment was placed inside a camera with D65 illuminant to evaluate the appearance. 212 Data acquisition was performed using Compusense® five release 5.0 (Compusense Inc., 213 Guelph, Ontario, Canada).

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215 2.7 Statistical analysis

Effect of reduction/replacement of NaCl and processing time on the variables studied 216 217 (chemical and microbial) was done by a two-factor analysis of variance (ANOVA) using the statistic software XLSTAT 2009.4.03 (Addinsoft, Barcelona, Spain). Fisher test was used to 218 evaluate differences among treatments. The effect of reduction/replacement of NaCl on texture, 219 220 sensory parameters and volatile compounds at the end of the process was done by one factor 221 ANOVA analysis. Furthermore, principal component analysis (PCA) was done to evaluate the 222 relationships among sausages and different parameters (pH, TBARS, ions, texture parameters, 223 moisture, lipids and protein content and aroma active compounds).

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225 **3. Results**

3.1. Chemical and microbiology analyses

At the end of the ripening process, the three treatments showed weight losses of 38.9-39.2 % (data not shown), that are suitable values for this kind of sausage. Salt reduction did not produce differences in weight losses among treatments at the end of process, as also observed other authors (Campagnol et al., 2011). 231 Two essential factors such as a_w and pH, guarantee the stability and safety of the 232 sausage. The pH and a_w values are shown in Fig. 1A. The pH dropped to 4.5 due to the LAB 233 growth in the treatments and then, pH experienced a slight increase due to ammonia formation. 234 No differences in pH were observed among treatments as also was observed in salt reduced 235 small caliber fermented sausages (Gelabert et al., 2003; Gou et al., 1996). Concerning aw, it decreased throughout the processing to 0.92 value in all treatments. At 9 days, differences 236 were found in a_w as seen by a highest a_w in the RS treatment, nevertheless the final sausages 237 did not show differences (p>0.05) as also has been reported by other authors (Campagnol et 238 239 al., 2011). However, Olesen et al. (2004) also observed a lowest aw in highly salted sausages 240 as we detected, although they performed a higher salt reduction in their assays (50 %) than the reduction done in our study (16 %). 241

The number of LAB and Staphylococci were within the range of what could be expected in dry fermented sausages and no differences were observed (p>0.05) among treatments throughout the process (data not shown). The population of LAB experienced a growth of 3 logarithmic cycles during the first 9 days and it was maintained stable until the end of the process. The number of *Staphylococci* suffered a slight decrease of 2 logarithmic cycles during processing.

In relation to fat and protein content, an increase in both contents was observed as a result of the reduction in moisture content during ripening. Salt content did not cause significant differences (p>0.05) in chemical composition among treatments at the end of the process (Table 1).

The color of sausages was also measured along the process, obtaining L*, a* and b* coordinates (data not shown). The trend in the three color coordinates throughout the process was similar to that observed by Olivares et al. (2010). No differences were detected in the final product among treatments as also has been observed in similar fermented sausages (Campagnol et al., 2011).

TBARS values increased during the drying process in the three treatments (Fig. 1B) as 257 it has been reported in similar sausages (Olivares et al., 2011). A highest oxidation was 258 259 observed in the reduced salt treatments at 9 days. TBARS values were significantly higher in 260 the RS treatment than S treatment, but at the end of the process only RSK was significantly higher than S treatment. The effect of NaCl on lipid oxidation is not clear. Several authors have 261 reported a pro-oxidant effect of NaCl in meat and meat products (Kanner, Harel & Jaffe, 1991; 262 263 Shahidi, Rubin & Wood, 1988) while other authors have not observed this effect in model 264 system (Sárraga & García-Regueiro, 1998). Nevertheless, Zanardi et al. (2010) also observed a highest oxidation in reduced salt sausages and they attributed this highest oxidation to the 265 266 use of CaCl₂ that can favored the lipid oxidation. In our study, the highest oxidation observed in 267 RSK treatment could be due to a slightly highest fat content observed in this treatment (RSK).

268 As expected, a significant reduction (p<0.05) of the Na⁺ ion content was achieved in RS 269 and RSK treatments throughout the process (data at the end of the process 57d are shown in 270 Table 1), however, at the end of the process only significant differences were found between S and RSK. The content of K^+ ion was increased (p<0.05) in RSK treatment, since this treatment 271 272 was the one containing KCI (Table 1). Finally, significant differences were detected in Cl ion content throughout the process among treatments although at the end of the process there 273 were not significant. The salt reduction detected in the treatments (RS and RSK) could be 274 considered as a healthy benefit, following EU indications. 275

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3.2. Texture profile analysis

278 Sausages TPA parameters were analyzed at the end of ripening and are shown in Table 279 2. Salt content did not produced differences in hardness, adhesiveness and springiness. 280 However, reduced/replaced treatments presented lower significant values of cohesiveness and 281 consequently chewiness, since this second parameter is the product of hardness, cohesiveness 282 and springiness. It is well known that salt favors the gel formation in fermented sausages and 283 leads to the desirable texture (Ruusunen, M., & Puolanne, E. 2005). However, we have 284 obtained that in slow fermented sausages the reduction/substitution of low salt percentages can 285 affect the cohesiveness and chewiness although the hardness is not affected. Therefore, it is necessary to determine if these changes can be detected by consumers. Only few studies have 286 287 detected differences in texture parameters by TPA analyses when the level of salt substitution was 40% or higher but not in lower percentages of substitution as we have performed. In this 288 sense, Gou et al. (1996) did not detect differences in texture parameters when KCI was used as 289 290 unique salt substitute while only Gimeno et al. (1999) reported a decrease in sausage hardness 291 when KCI was used in combination with other divalent salts.

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293 **3.3. Sensory analysis**

The results of sensory analysis are shown in Table 3. The sensory panel did not detected significant differences among treatments in appearance and tenderness acceptability, however, S treatment had the highest acceptability in aroma, taste, juiciness and overall quality.

297 Previous studies detected differences in sensory texture parameters when the level of 298 salt substitution was 40% or higher but not in lower percentages of substitution as we have performed. In this sense, Gou et al. (1996), Gelabert et al. (2003) and Guardia et al. (2008) 299 300 reported an increase in hardness when KCI was used as unique salt substitute. However, the 301 use of KCI in combination with other divalent salts or lactate resulted in a decrease in sausage 302 hardness (Gimeno et al., 1999; Gelabert et al., 2003). Also, it is important to remark that these 303 previous studies were performed mainly in small diameter fermented sausages and there are 304 no reports in slow fermented sausages. Nevertheless, only the juiciness acceptability was the 305 texture parameter that the consumers detected as lowest in reduced salt treatment (RS). The

lowest juiciness acceptability detected in RS treatment could be due to the lower cohesivenessand chewiness (Table 2) observed in this treatment.

It was remarkable to observe that the addition of KCI removed the differences observed in taste, juiciness and overall quality and the consumers showed the same acceptance between S and RSK treatments in these parameters. Therefore a 16 % substitution by KCI can be carried out although the aroma was the unique parameter that was not improved by KCI addition. Therefore, it is necessary to understand which aroma compounds are affected by the salt reduction and substitution.

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315 **3.4. Aroma compound analyses**

In order to study how salt reduction and substitution affects aroma development in slow 316 317 fermented sausages, the volatile compounds were extracted by SPME and analyzed by GC-MS and olfactometry analysis (Table 4 and 5, respectively). It is necessary to take into account that 318 319 the proportion of volatile compounds depends on the extraction method used. In the present 320 study, SPME technique with CAR/PDMS fiber was used. A total of 96 compounds were 321 identified at the end of the process (Table 4) being 20 aldehydes, 11 alkanes, 13 ketones, 1 322 pyrazynes, 8 sulfur compounds, 8 acids, 17 alcohol, 10 esters, 6 aromatic hydrocarbons and 1 323 terpene. The volatile compounds present in the sausage treatments classified by chemical groups are shown in Figure 2. The reduction and substitution of salt produced an increase in 324 325 aldehyde compounds but also a reduction in the abundance of sulphur and acid compounds (Figure 2). One of the chemical groups that was present in highest abundance in the three 326 treatments were the acids, representing 61-72 % of the total extracted area, followed by 327 328 aldehydes (8-16%) and alcohols (5-9%). Acetic acid was the most abundant compound in 329 headspace (HS) (Table 4). Other abundant compounds were hexanal, 3-methyl-thiophene, 330 octanoic acid, phenol, octane, heptanal, 1-hexanol, hexanoic acid, nonanal, octanal, pentanal, 2-butanone and 3-methyl-2-butenal (Table 4). All the identified compounds have been 331 previously reported in fermented sausages (Marco, Navarro & Flores, 2004, 2006, 2008; 332 Olivares et al., 2011) using the same extraction technique except 3-methyl-2-butenal, 2-hydroxy 333 334 benzaldehyde and butyl acetate. The identification of compounds was confirmed with authentic 335 standards except benzyl alcohol and methyl 2,4-hexadienoate which were tentatively identified.

In order to study the effect of salt reduction/substitution on the processes involved in the generation of aroma compounds is better to classify the volatile compounds according to their possible origin: lipid autooxidation, bacterial metabolism (lipid β -oxidation, carbohydrate fermentation, amino acid degradation and Staphylococci esterase activity) and unknown or contaminant compounds (Table 4) as indicated Ordoñez, Hierro, Bruna and de la Hoz (1999).

The carbohydrate fermentation volatile compounds were the most abundant compounds, representing 53-58 % of the total extracted area, since acetic acid just represented a 44-52 %. Then, lipid autooxidation volatile compounds represented 16-24 %, amino acid degradation products 8-10 %, volatile compounds derived from staphylococci esterase activity
0.6-1 % and lipid β-oxidation products 0.6-0.8 %.

Volatile compounds derived from lipid autooxidation have an important role in the odor 346 of dry fermented sausages due to their low olfactory threshold (Marco et al., 2007). 347 Predominantly, the lipid oxidation originates aldehydes among other products such as alkanes, 348 ketones, alcohols, etc. Salt content affected (p<0.05) the HS abundance of volatile compounds 349 as observed by a highest abundance in RS and RSK treatments (Table 4). Several compounds 350 351 have a significant higher abundance in RSK treatment than S and RS treatments such as 352 hexanal, butanal, 2-ethylfuran, 1-pentanol and 2-octenal (Table 4). However, only 2-pentylfuran 353 was more abundant in the HS of RSK treatment than in S treatment while tridecane showed opposite effect. Only two compounds, 1-octanol and 1-heptene, showed more abundance in the 354 HS of RS and RSK than S treatment while hexane and octanoic acid displayed the opposite 355 356 effect. Finally, octane and heptanoic acid and 1-propanol showed a greater abundance in HS of 357 RS than S and RSK treatments. The higher abundance of compounds derived from lipid oxidation in RS and RSK treatments is in accordance to the TBARS values obtained as they 358 359 were significantly higher in RS and RSK treatments, probably due to their highest fat content.

360 On the other hand, salt content did not produced significant differences on volatile 361 compound derived from lipid β -oxidation reactions except for 2-nonanone which was more abundant in the HS of S treatment than in reduced/replaced treatments (RS, RSK). However, 362 363 several compounds derived from carbohydrate fermentation were affected by salt reduction. 364 The carbohydrate fermentation reactions mainly generate acids, followed by alcohols and ketones. Only ethanol, 2,3-butanediol and 2,3-butanedione showed significant differences 365 among treatments (Table 4). 2,3-butanediol and 2,3-butanedione showed a greater HS 366 367 abundance (p<0.05) in the S treatment while ethanol had the lowest abundance in S treatment. The reduction in 2,3-butanediol abundance was also observed by Olesen et al. (2004), however 368 they detected the opposite effect for 2,3-butanedione. These authors related the 2,3-369 370 butanedione concentration to the activity of Staphylococcus starter although in the present 371 study we did not detect differences in Staphylococci grow among the treatments.

372 Volatile compounds derived from amino acid degradation depend on free amino acid 373 concentration present in sausages. Branched chain amino acid produces branched aldehydes, 374 alcohols and acids; in addition sulfur amino acids generate sulfur volatile compounds as well as 375 aromatic amino acids produces aromatic compounds. Salt affected the total abundance of this 376 group of compounds as a highest HS abundance (p < 0.05) was detected in S than RS 377 treatment. The most abundant compound in HS of sausages within this chemical family was 3-378 methyl thiophene (Table 4). This compound had a higher abundance in S treatment than RS 379 and RSK treatments, and dimethyl disulfide was also more abundant in S treatment, but it was only significantly different from RSK. Nevertheless, ethyl methyl sulfide showed the lowest 380 381 abundance in RSK treatment while 3-methyl 2-butenal showed the opposite effect. On the other hand, several compounds, 3-methyl thiopropanal and benzaldehyde, showed a greater abundance in the HS of reduced treatments (RS, RSK) than in S treatment. However, phenylethyl alcohol and benzeneacetaldehyde were only more abundant in the HS of RS treatment (Table 4).

386 The compounds derived from the Staphylococci activity were also affected by the salt reduction. A higher abundance of total ester compounds was found in the HS of RS treatment, 387 but due to variability among sausages, the differences were not significant. However, several 388 compounds were significantly different among treatments, ethyl acetate, ethyl butanoate, ethyl 389 2-hydroxy propanoate and ethyl hexanoate were significantly higher in RS treatment than S and 390 391 RSK treatments (Table 4). These ester compounds provides fruity notes and have been widely 392 detected in slow fermented sausages (Olivares et al., 2011). Talon, Chastagnac, Vergnais, 393 Montel & Berdagué (1998) indicated that the production of esters compounds depended on the 394 presence of the substrates (ethanol and acids) and on the Staphylococci esterase activity. In 395 the present study we detected highest ethanol abundance in RS treatment followed by RSK treatment and S treatment that would explain the highest ester production found at the same 396 397 proportion in the treatments.

About unknown or contaminants compounds, few differences were found among treatments. The presence of sorbic acid and its ethyl and methyl esters came from the potassium sorbate applied to the sausage casing to avoid mold growth as also reported Olivares et al. (2011). Sorbic acid and their esters showed highest abundance in RS treatment while dimethyl sulfone had a highest abundance RSK treatment.

403 Generally, all the studies performed on volatile compounds in reduced fermented 404 sausages studied a percentage reduction of 25 % or higher (Olesen et al., 2004; Ravyts et al., 405 2010; Campagnol et al., 2011). While Campagnol et al. (2011) reported few changes in the profile of volatile compounds when salt was substituted in 25 and 50 % by KCI, other authors 406 such as Olesen et al. (2004) indicated a considerable impact on the volatile profile when NaCI 407 was reduced in a 50 % in fermented sausages. In addition, Olesen et al., (2004) indicated that 408 a half percent salt reduction produced an activation of lactic acid bacteria growth giving a higher 409 410 pH drop affecting negatively the growth of Staphylococci. This effect produced a decrease in 411 the generation of branched derived volatile compounds in low salted sausages. However these authors reported that the differences observed among high and low salted sausages were 412 narrowed as the ripening process continued. In the present study, we only analyzed a reduction 413 414 of the 16 % percent of the total salt content without observing any effect on the growth of lactic 415 acid bacteria and Staphylococci, also no differences in pH were detected along the process 416 among treatments. Therefore, the differences that we have observed in volatile compounds 417 derived from amino acid degradation cannot be attributed to a higher Staphyloccoci activity. 418 This fact is also in accordance to Ravyts et al. (2010) who indicated that modifications of salt 419 concentrations had a limited impact on the growth of sausage microbiota and they did not find a

significant effect on volatile production. Probably these authors did not find an effect on volatile
compounds production because they only extracted few volatile compounds as they used static
headspace gas chromatography analysis. However, it is necessary to remember that the
flavour of fermented sausages is affected by recipe and the type of starter culture used (Leroy,
Verluyten & De Vuyst, 2006).

In order to reveal the aroma contribution of the volatile compounds present in the slow 425 426 fermented sausages an olfactometry analysis was performed showing the presence of thirty-427 five different aroma active zones. Twenty six of them were identified by mass spectra, linear retention indices and odor description, while nine of them could not be identified (Table 5). All of 428 429 identified compounds have been previously detected as aroma impact compounds in fermented sausages (Chevance, Farmer, Desmond, Novelli, Troy & Chizzolini, 2000; Gianelli, 430 Olivares & Flores, 2011; Marco et al., 2007; Meynier, Novelli, Chizzolini, Zanardi & Gandemer, 431 432 1999; Olivares et al., 2011; Schmidt & Berger, 1998a, 1998b; Söllner & Schierberle, 2009; 433 Stahnke, 1995a, 1995b) except 2-hydroxy benzaldehyde (herbal, stable, roasted bread), butyl 434 acetate (spice, rancid, wood, boiled vegetables) and 3-methyl thiophene (cooked potato, green, 435 wood). The detection frequency (DF) method was applied to determine the contribution of the 436 different volatile compounds to the aroma of slow fermented sausages. The highest DF values 437 mean a highest aroma impact. The most potent odorants detected were 2-hexenal (roasted, meat broth), 1-octen-3-ol (mushroom), acetic and butanoic acids (vinegar and cheese odors, 438 439 respectively), dimethyl trisulfide (onion, cabbage), 2-nonanone (plastic, wood), 3-methyl thiopropanal (cooked potato, savory) and D-limonene (citrus). Four of these aroma compounds 440 (acetic and butanoic acids, 1-octen-3-ol and 3-methyl thiopropanal) were also detected as 441 potent odorants in fermented sausages (Olivares et al., 2011). However other potent aroma 442 443 compounds, contributing with roasted nuts odors, were detected although they were not identified (unknown compounds with LRI 1179 and 1223). These last unknown aroma 444 compounds were also detected in similar fermented sausages by Olivares et al. (2011) using 445 446 the same extraction technique.

In order to study which aroma compounds were responsible for the highest acceptability 447 448 of the salted treatment (S), a principal component analysis was done using the following parameters: chemical composition (fat, protein and moisture content), pH, ions (Na⁺, K⁺, Cl⁻), 449 aroma active volatile compounds (those shown in Table 5) and texture parameters. Figure 3 450 illustrates the results of the PCA analysis. Two principal components were able to explain the 451 452 57.88 % of the total variance observed. PC1 is the most important variable because it 453 accounted for 39.83 % of the variance while PC2 accounted for 18.05 % of the variance. PC1 differentiated the sausages by their salt content. S treatment, with the highest salt content, 454 455 appeared separately in the positive part of PC1, associated with higher texture parameters, 456 higher Na+ and CI- ions content and the presence of the volatile compounds such as 2nonanone, dimethyl trisulfide, 3-methyl thiophene, 2,3 butanedione and acetic acid. However, 457

PC2 differentiated samples with reduced salt content (RS) to S and RSK treatments showing a
negative correlation with RS treatment. Therefore, sausage samples with KCI as substitute
(RSK treatments) were differentiated but more similar to the S treatment than the reduced salt
treatment (RS).

462

463 4. Conclusion

In summary, small salt reduction (16%) affected the quality of slow fermented sausages 464 producing a reduction in the acceptance of aroma, taste, juiciness and overall quality. However, 465 the substitution by KCI removed the differences observed in taste, juiciness and overall quality 466 467 and the consumers showed the same acceptance for high salt (S) and substituted (RSK) treatments. Therefore a 16 % substitution by KCl can be carried out however, the aroma was 468 469 the unique parameter that was not improved by KCI addition. The aroma perceived by 470 consumers was affected due to the reduction detected in sulfur and acid compounds and the 471 increase in aldehyde compounds. Moreover, the aroma compounds that characterized the high salt treatment (S) were dimethyl trisulfide, 3-methyl thiophene, 2,3-butanedione, 2-nonanone 472 473 and acetic acid. In addition, the decrease in chewiness and cohesiveness detected in reduced 474 and substituted treatments (RS and RSK) could affect the perception of the aroma compounds. 475 To improve the aroma of reduced salt slow fermented sausages is necessary to look for other 476 alternatives to KCI addition to improve the aroma perception. Further studies about the use of 477 salt-associated odours which can induce a saltiness enhancement should be performed.

478

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480

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484

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Figure legends

Figure 1. Changes in pH, Aw (A) and TBARS (B) during the ripening of dry fermented sausages: S (control, ○), RS (16 % reduced salt, ∇) and RSK (16% KCl to replace NaCl, □)

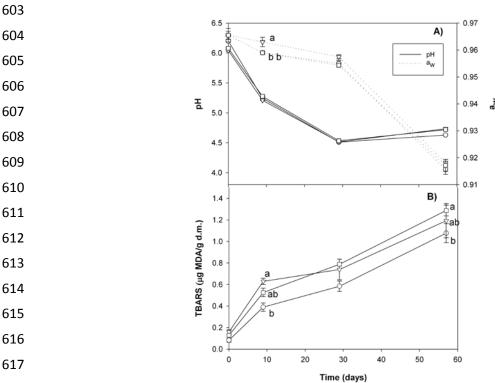


Figure 2. Total volatile compounds abundance expressed as AU x 106 in the headspace of dry fermented sausages with different salt content: S (control), RS (16 % reduced salt) and RSK (16% KCl to replace NaCl) at the end of the ripening process. Different letters in the same chemical group indicate significant differences (p<0.05) among treatments.

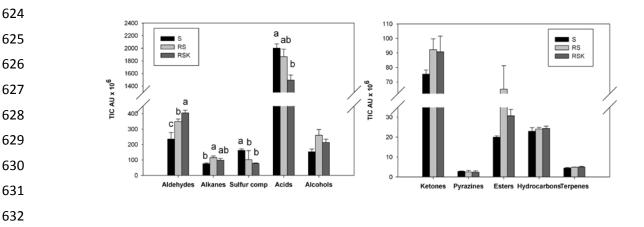
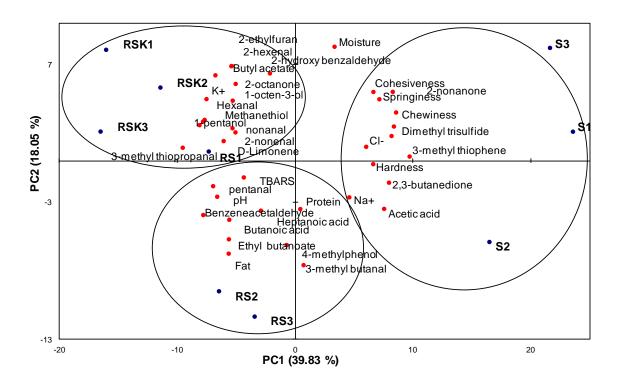


Figure 3. Loadings of the first two principal components (PC1-PC2) of the analyzed parameters
(pH, TBARS, ions, texture parameters and aromatic active compounds) of fermented sausages
with different salt content: S (control), RS (16 % reduced salt) and RSK (16% KCI to replace
NaCI) at the end of the ripening process.



		S		RS (red16%)		RSK (red16% +KCI)	Р
Moisture	(%)	49.51		48.06		49.23	ns
Fat	(%)	10.75		12.73		12.32	ns
Protein	(%)	35.32ab		37.00		34.88b	ns
Na⁺	(g/100 g d.m.)	3435.38	а	3074.55	ab	2748.67 b) ***
K⁺	(g/100 g d.m.)	952.26	b	996.35	b	1458.14 a	***
Cl	(g/100 g d.m.)	3257.52		2946.78		2992.77	ns

Table 1. Chemical composition and ion contents in dry fermented sausages with different salt content: S (control), RS (16 % reduced salt) and RSK (16% KCI to replace NaCI) at the end of the ripening process (57 days).

P: P value of salt content effect. *** P<0.001, ** P<0.01, * P<0.05, ns: P>0.05. Identical letters in each row indicate the absence of differences at p>0.05 (Fisher test).

Table 2. Texture parameters of dry fermented sausages with different salt content: S (control), RS
(16 % reduced salt) and RSK (16% KCI to replace NaCI).

	S	RS (red16%)		RSK (red16% +KCl)	5 F	Ρ
Hardness (N)	257.85	245.43		246.23	r	าร
	(22,60)	(17.22)		(14.64)		
Adhesiveness (N·s)	-3.34	-3.26		-3.68	r	าร
	(0.54)	(0.47)		(0.58)		
Springiness	0.63	0.61		0.61	r	าร
	(0.02)	(0.03)		(0.03)		
Cohesiveness	0.64 a	a 0.62	b	0.62 b		*
	(0.02)	(0.02)		(0.01)		
Chewiness	103.49 a	a 93.11	b	93.24 b		**
	(9.21)	(8.96)		(7.94)		

P_S: P value of salt content effect. *** P<0.001, ** P<0.01, * P<0.05, ns: P>0.05. Identical letters in each row indicate the absence of differences at p>0.05 (Fisher test). The values represent the mean and (standard deviation).

Table 3. Sensory acceptability of dry fermented sausages with different salt content: S (control), RS (16 % reduced salt) and RSK (16% KCI to replace NaCI) at the end of the ripening process.

	S	RS (red16%)	RSK (red16% +KCI)	Р
Appearance	6.14	5.86	5.96	ns
	(1.57)	(1.41)	(1.48)	
Aroma	6.33 a	5.93 b	5.89 b	*
	(1.46)	(1.37)	(1.44)	
Taste	5.96 a	5.34 b	5.84 a	**
	(1.89)	(1.80)	(1.94)	
Tenderness	6.13	6.00	6.18	ns
	(1.64)	(1.69)	(1.61)	
Juiciness	6.22 a	5.75 b	5.99 ab	*
	(1.63)	(1.60)	(1.58)	
Overall quality	5.92 a	5.54 b	5.92 a	*
	(1.72)	(1.61)	(1.78)	

P_s: P value of salt content effect. *** P<0.001, ** P<0.01, * P<0.05, ns: P>0.05. Identical letters in each row indicate the absence of differences at p>0.05 (Fisher test). The values represent the mean and (standard deviation).

Table 4. Volatile compounds (expressed as AU x 10^6 extracted by HS-SPME) identified in the headspace of dry fermented sausages with different salt content: S (control), RS (16 % reduced salt) and RSK (16% KCl to replace NaCl) at the end of the ripening process.

		Sausage Batches								
Compound	LRI ^B	RI ^C	S		RS (red16%)		RSK (red16% +KCl)		SEM	P
Lipid autooxidation										
Pentane	500	а	8.25	С	19.98	b	32.45	а	1.66	***
Propanal	524	а	4.78		11.00		9.78		2.62	ns
Isopropyl alcohol	542	а	3.26		6.58		10.41		1.92	ns
Hexane	600	а	12.05		5.81		5.88		1.32	*
1-Propanol (31) ^A	613	а	0.06	b	0.11	а	0.09	ab	0.01	*
2-Methylfuran (82) ^A	615	а	1.23		1.21		0.69		0.15	ns
Butanal	622	а	0.56	b	1.41	b	3.04	а	0.27	**
1-heptene (55) ^A	693	а	0.03	b	0.25	а	0.31	а	0.03	**
Heptane $(71)^{A}$	700	а	1.84		5.55		5.31		1.19	
2-Ethylfuran (81) ^A	720	а	1.55	b	1.13	b	3.11	а	0.32	*
Pentanal (44) ^A	737	а	18.82	ь.	33.63	_	33.67		5.19	ns *
Octane	800	a	25.95	D	60.93	а	37.00	D	5.21	
2-octene 1-Pentanol	810	a	13.15	h	8.91	h	6.13	~	1.66	ns **
Hexanal (44) ^A	826 840	a a	9.35 72.90		12.22 92.41		17.79 137.28		1.11 10.45	*
Nonane	900	a a	3.14	D	2.56	D	2.36	a	0.28	
2-Hexenal (Z)	900 904	a a	1.84		1.47		2.56		0.20	ns
2-Butylfuran	904	a	1.71		2.06		1.88		0.35	ns
1-Hexanol	922	a	28.19		45.87		37.14		5.18	ns
Heptanal	940	a	19.06		54.79		55.22		9.72	
Decane	1000	a	1.81		1.84		1.85		0.20	ns
2-Penthylfuran	1009	a	3.61	b	8.73	ab	11.78	а	1.51	*
Octanal	1047		19.79	Ň	31.80	uo	33.30	ŭ	3.36	ns
Hexanoic acid	1075	a	37.54		42.70		43.92		2.15	ns
2-Ethyl 1-hexanol	1082	а	6.89		7.54		8.00		0.60	ns
Undecane (57) ^A	1100	а	0.16		0.15		0.18		0.02	ns
2-Octenal (Z)	1115	а	1.03	b	1.88	b	4.24	а	0.64	*
1-Octanol	1123	а	1.02	b	2.84	а	2.74	а	0.40	*
Nonanal	1149	а	28.86		37.94		37.95		3.62	ns
Heptanoic acid	1165	а	2.03	b	3.23	а	2.16	b	0.29	*
Dodecane	1200	а	6.32		7.19		5.36		0.62	ns
2-Nonenal (Z)	1221	а	2.12		3.09		2.90		0.39	ns
Decanal	1256	а	1.99		2.36		1.73		0.35	ns
Octanoic acid	1266	а	87.82	а	35.80	b	36.79	b	5.78	***
2,4-Nonadienal (E, E)	1287	а	tr.		tr.		tr.			
Tridecane	1300	а	2.83	а	2.05	ab	1.39	b	0.24	*
Nonanoic acid	1357	а	3.20		2.83		2.85		0.16	
Decanoic acid	1449	а	7.98	ь.	6.80	_	6.97	_	0.90	
Total			442.70	b	566.66	а	606.23	а	36.77	^
Bacterial metabolism										
Lipid β oxidation	744	_	0.47		0.4.4		0.00		0.00	
2,3-Pentanedione (85) ^A	744		0.17		0.14		0.08		0.03	
2-Heptanone	933		4.64		7.88		7.70		1.10	
2-Heptanol 1-Octen-3-ol (57) ^A	946 1030		5.92 2.13		4.32 2.01		4.01		0.67 0.56	
2-Octanone	1030		0.62		2.01		3.80		0.56	
2-Octanone 2-Nonanone	1039		0.62 5.60	2	3.01	h	0.90 3.78	h	0.09	ns **
2-Undecanone	1306		1.27	a	1.25	U	1.11	U	0.29	
Total	1300	a	20.36		19.22		21.38		2.20	
Carbohydrate fermentation			20.00		13.22		21.00		2.20	113
Acetaldehyde	466	а	9.94		12.44		10.67		0.67	ns
Ethanol	400 508		9.94 13.11	~	96.86	2	49.08	h	12.11	115 **
Acetone	530		20.48	U	90.00 27.79	a	49.08 24.48	U	4.18	
2,3-Butanedione (43) ^A	530 626		20.48	2	0.34	h	24.46	h	4.18 0.04	115
	020	u	0.02	a	0.04	5	0.00	5	0.04	ns

Acetic acid 737 a 1447.65 1295.19 1228.50 59.09 ns Butanoic acid 892 a 72.74 80.23 82.45 4.31 ns Total 1595.03 1544.06 1424.30 75.20 ns Amino acid degradation 1595.03 1544.06 1424.30 7.520 ns 2-Methyl propanal 594 a 7.42 7.38 5.59 1.06 Ns Ethyl methyl sulide (61) ^h 624 a 0.85 1.00 1.20.8 Ns 2-Methyl butanal (44) ^h 689 a 8.89 9.32 8.36 0.96 Ns 2-Methyl butanal (58) ^h 700 a 2.50 2.30 2.11 0.21 Ns 3-Methyl-3-buten-1 col (41) ^h 789 a 0.30 0.24 0.29 0.05 Ns 3-Methyl-butenal 650 a 6.624 b 9.89 1.166 a 0.75 * 3-Methyl-butenal 5.16									
Butanoic acid 892 a 72.74 80.23 82.43 4.31 ns Total 1595.03 1544.06 1424.30 75.20 ns Amino acid degradation 2.Methyl propanal 594 a 7.42 7.38 5.59 1.06 Ns Ethyl methyl sulfide (61) ^h 624 a 0.98 a 0.79 a 0.41 b 0.10 * Benzene 675 a 0.95 a 1.00 1.20 0.13 Ns Ns 2-Methyl hytopanal 682 a 0.36 0.20 0.31 0.08 Ns Ns 3-Methyl butanal (58) ^h 700 a 2.50 2.30 2.31 0.021 Ns 0.21 Ns 3-Methyl butanal (55) ^h 84 0.33 14.11 1 13.63 0.76 Ns Ns 3-Methyl hiophene 794 a 125.16 a 68.25 b 42.66 b 6.88 *** 3-Methyl hiopropanal 966 a 6.24 b 9.89 a 11.66 a 0.75 *** Ethyl benzene 883 a 2.46 b 1.85 a 1.66 a 0.55 *** 3-Methyl hiopropanal 966 a 6.24 b 9.89 a 11.6 a 0.20 Ns									
Total 1595.03 1544.06 1424.30 75.20 ns Amino acid degradation 594 a 7.42 7.38 5.59 1.06 Ns Ethyl methyl sulfide (61) ^h 624 a 0.98 a 0.79 a 0.41 b 0.10 * 2-Methyl butanal (44) ^h 682 a 0.36 0.20 0.31 0.08 Ns 3-Methyl butanal (44) ^h 689 a 8.89 9.32 2.36 0.96 Ns 2-Methyl hutanal (44) ^h 689 a 2.40 2.30 2.11 0.21 Ns 3-Methyl hutanal (45) ^h 772 a 4.40 a 3.11 a 2.66 Ns 3-Methyl hophene 794 a 12.83 14.11 13.63 0.76 Ns 3-Methyl hophene 794 a 2.61 2.53 2.42 0.52 Ns 3-Methyl hyd przine 944 a 2.81 2.53 2.42					а		b		
Amino acid degradation 2-Methyl propanal 594 a 7.42 7.38 5.59 1.06 Ns Ethyl methyl sulide (61) ^h 624 a 0.98 a 0.79 a 0.41 b 0.10 * Eenzene 675 a 0.95 1.00 1.20 0.13 Nos Nos Nos Nos Nos Nos Nos 0.41 b 0.01 * 2-Methyl butanal (44) ^h 689 a 8.89 9.32 8.36 0.96 Ns 2-Methyl butanal (55) ^h 700 a 2.50 2.30 2.11 0.21 Ns 3-Methyl-3-buten-1-ol (41) ^h 778 a 125.61 a 6.82.5 b 4.88 * 4.6 2.74 2.90 0.20 Ns 3-Methyl-3-buten-1-ol (41) ^h 778 a 2.61 * 1.01 0.27 2.24 0.22 Ns 3.4 2-5-dimethyl prozinal 3 2		892	а						
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2-Methyl butanal (58) ^A 700 a 2.50 2.30 2.11 0.21 Ns Dimethyl disulfide 772 a 4.40 a 3.11 ab 2.06 b 0.51 * Toluene 788 a 12.83 14.11 13.63 0.76 Ns 3-Methyl hiophene 794 a 125.16 a 68.25 b 42.68 b 6.88 *** 3-Methyl Jebutanal (55) ^A 840 a 14.73 b 17.74 b 28.74 a 2.61 * Ethyl benzene 833 a 2.46 2.74 2.90 0.20 Ns 3-Methyl thiopropanal 966 a 6.24 b 9.89 a 11.66 a 0.75 ** Dimethyl trisulfide 1002 a 1.80 1.25 1.01 0.20 Ns Benzaldehyde 1107 a 16.33 b 21.55 a 22.29 a 1.40 * 2-Hydroxy benzaldehyde (122) ^A 1100 a 0.27 0.27 0.27 0.28 b 0.01 Ns Benzaldehyde 1107 a 3.69 c 6.82 a 5.65 b 0.30 *** Phenol 1111 a 75.55 74.23 73.78 2.99 ns Benzyl alcohol 1120 b 1.63 1.66 1.53 0.05 ns Benzyl alcohol (91) ^A 1194 a 0.39 b 2.32 a 0.92 b 0.16 *** Total 2.89,69 a 247.69 b 227.82 ab 16.78 ns Stephylococci esterasa activity Methyl acetate 551 a 2.20 a 1.30 *** Bethyl acetate 635 a 6.76 b 26.99 a 13.21 b 2.58 ** Bthyl acetate 635 a 6.76 b 26.99 a 13.21 b 2.58 ** Bthyl acetate 635 a 0.76 b 26.99 a 13.21 b 2.58 ** Bthyl acetate 635 a 0.77 0.62 0.47 0.12 ns Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.44 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.44 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.44 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.44 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.30 ** Bthyl 2-hydroxy-propanoate 869 a 1.78 b 16.45 a 1.97 b 0.09 *** Bthyl acetate 637 a 1.78 b 16.45 a 1.97 b 0.09 *** Bthyl acetate 647 a 2.73 a 4.42 3.14 0.08 ns Carbon disulfide 372 a 3.73 a 0.23 ns Styrene 918 a 1.61 1.27 1.36 0.15 ns Divendence 1023 a 5.07 7.15 6.49 0.82 ns DyCymene 916 a 3.24 3.23 3.33 0.23 ns Styrene 916 a 3			а	0.36					0.08 Ns
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3-Methyl thiopropanal 966 a 6.24 b 9.89 a 11.66 a 0.75 ** Dimethyl trisulfide 1002 a 1.80 1.25 1.01 0.20 Ns Benzaldehyde 1017 a 16.33 b 21.55 a 2.229 a 1.04 * 2-Hydroxy benzaldehyde (122) ^A 1100 a 0.27 0.27 0.28 0.01 Ns Benzeneacetaldehyde 1107 a 3.69 c 6.82 a 5.65 b 0.30 **** Phenol 1120 b 1.63 1.66 1.53 0.05 ns Benzy lalcohol (91) ^A 1194 a 0.39 b 2.32 a 0.92 b 0.16 **** Total 27.62 a 16.78 ns Staphylococci esteras activity staphylococci (91/ ^A 311 0.27 0.62 0.47 0.12 ns Ethyl acetate 551 a 6.76 2.99 a 1.29 0.30	Ethyl benzene	883	а					2.90	
$\begin{array}{ccccccc} 0.000 & 0$	2,5-dimethyl pyrazine	-	а	2.81		2.53		2.42	
Benzaldehyde1017a16.33b21.55a22.29a1.40*2-Hydroxy benzaldehyde1100a0.270.280.01NsBenzeneacetaldehyde1107a3.69c6.82a5.65b0.30****Phenol1111a75.5574.2373.782.99nsBenzyl alcohol1120b1.631.661.530.05nsPhenylethyl alcohol (91)^A1194a0.39b2.32a0.92b1.678Total289.69a247.69b227.82ab16.78nsStaphylococci esterasa activityMethyl acetate635a6.76b26.99a1.321b2.58***Ethyl acetate635a6.76b26.99a1.29b0.04nsEthyl butanoate (71)^A831a0.28c2.90a1.29b0.04***Butyl acetate847a2.052.554.690.64nsssfthyl acetate1028a0.10c1.22a0.71b0.09***Ethyl acetate1229a2.734.423.140.83nsnssfthyl acetate3.630.10sssssssssssssss<					b		а		0.75
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Staphylococci esterasa activityMethyl acetate551a2.701.981.620.27nsEthyl acetate635a6.76b26.99a13.21b2.58**Ethyl popanoate $(57)^A$ 744a0.270.620.470.12nsEthyl butanoate $(71)^A$ 831a0.28c2.90a1.29b0.30**Butyl acetate847a2.052.554.690.64ns***Ethyl butanoate $(88)^A$ 1028a0.10c1.22a0.71b0.09***Ethyl octanoate(88)^A1028a0.10c1.22a0.71b0.09***Ethyl octanoate1229a2.734.423.140.83nsrsTotal16.6857.1227.099.78nsUnknown or contaminants compound472a7.839.179.840.47nsCarbon disulfide537a13.787.758.531.59nso-Xylene916a3.243.233.330.23nsStyrene918a1.611.271.360.15ns2-Butoxyetanol952a2.352.742.870.31nsDutyrolactone1023a5.077.156.490.82nsp-Cymene1050a <t< td=""><td></td><td>1194</td><td>а</td><td></td><td></td><td></td><td></td><td></td><td>0.10</td></t<>		1194	а						0.10
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Total 387.29 445.65 136.87 82.35 ns					а		а		
		1195	а						
ALL: Abundance units the result of counting the total ion chromatogram (TIC) for each compound									

AU: Abundance units, the result of counting the total ion chromatogram (TIC) for each compound.

^A Target ion used to quantify the compound when the peak was not completely resolved.

^B Linear retention indices (LRI) of the compounds eluted from the GC-MS using a DB-624 capillary column (J&W Scientific 30 m x 0.25 mm i.d. x 1.4 μ m film thickness). ^c Reliability of identification: a, identification by mass spectrum and by coincidence with the LRI of an

authentic standard; b, tentatively identification by mass spectrum.

^D p value of salt content effect. ***: p<0.001, **: p<0.01, *: p<0.05, ns: p>0.05. Means followed by different letters in the same compound indicate significant differences (p<0.05) among batches.

Table 5. Odor active com	pounds identified in the HS of (dry fermented sausages

Compound	LRI LRI GC-O standard		Descriptor	DF	Previously reported in dry sausages	
Lipid autooxidation					· •	
2-Ethylfuran	725	718	Tallowy, savory, sweet	4	8	
Pentanal	734	735	Green, nut, meat broth	7	1,8	
1-Pentanol	815	820	Floral, butter, roasted nuts	5	8,10	
Hexanal	835	836	Fresh cut grass	8	1,2,5,7-9,11	
2-Hexenal (Z)	903	904	Sweet, roasted, meat broth	12	8	
Nonanal	1148	1151	Citrus, laurel, carnation	8	1,2,7-10	
Heptanoic acid	1163	1162	Herbal, rancid	6	8,10	
2-Nonenal (Z)	1219	1222	Herbal, strawberry	4	1,2,8-11	
2,4-Nonadienal (E, E)	1290	1288	Herbal, unpleasant, roasted	6	7,9,11	
Bacterial metabolism			, , , , , , , , , , , , , , , , , , , ,	-		
Lipid β oxidation						
1-Octen-3-ol	1024	1028	Mushroom	12	5,8,11	
2-Octanone	1031	1037	Green, garlic	10	11	
2-Nonanone	1137	1142	Plastic, wood, pop-corn, roasted	11	1,3,4,8	
Carbohydrate fermentation			·····, ····, ·····, ·········		, - , - , -	
2,3-Butanedione	632	632	Butter	4	1-4,6,8,10,11	
Acetic acid	705	700	Vinegar	11	1-3,8-11	
Butanoic acid	871	876	Cheese	12	1-4,8,9,11	
Amino acid degradation					, , ,	
3-Methyl butanal	691	691	Green	5	1,2,8,9,11	
3-Methyl thiophene	795	796	Cooked potato, green, wood	4	11	
3-Methyl thiopropanal	967	969	Cooked potato, savory	11	7-11	
Dimethyl trisulfide	1007	1009	Onion, rotten, cabbage	12	6	
2-Hydroxy benzaldehyde	1105	1107	Herbal, stable, roasted bread	4	-	
Benzeneacetaldehyde	1110	1112	Rancid, musk, jasmine	6	8-11	
Staphylococci esterasa activity						
Ethyl butanoate	825	825	Pineapple, strawberry	6	1-5,8-11	
Butyl acetate	843	840	Spice, rancid, wood, boiled vegetables	9	-	
Unknown or contaminants compo						
Methanethiol	472	471	Rotten, stable	8	8,11	
D-Limonene	1046	1048	Citrus	11	3-5,8	
4-Methyl-phenol	1194	1190	Plastic, stable, rancid,	7	3,4,9-11	
Unknown 1	922		Green, rancid, manure, cheese	6		
Unknown 2	962		Onion, Swiss chard	7		
Unknown 3	1001		Herbal, roasted, damp, vanilla	6		
Unknown 4	1179		Roasted nuts	10		
Unknown 5	1223		Roasted nuts, unpleasant, cardboard	11		

^A Linear retention indices (LRI) of the compounds or standards eluted from the GC-FID-O using a DB-624 capillary column (J&W Scientific 60 m x 0.32 mm i.d. x 1.8 μm film thickness).

^BDF Detection frequency value

^c Previously reported in dry fermented sausages by: 1 Stahnke (1994), 2 Stahnke (1995b), 3 Schmidt and Berger(1998a), 4 Schmidt and Berger (1998b), 5 Meynier et al. (1999), 6 Chevance et al. (2000), 7 Blank et al. (2001), 8 Marco et al. (2007), 9 Söllner et al. (2009), 10 Gianelli et al. (2011), 11 Olivares et al. (2011).