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Bio-guided fractionation of extracts of Geranium robertianum L.: Relationship between phenolic profile and biological activity

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ABSTRACT

Geranium robertianum L. is used in folk medicine and herbalism practice for the treatment of a variety of ailments. Recently, we studied the bioactivity of several aqueous and organic extracts of this plant. In this work, the more active extracts were fractionated and the fractions evaluated for their antioxidant activity and cytotoxicity against several human tumor cell lines and non-tumor porcine liver primary cells. Some of the fractions from the acetone extract consistently displayed low EC_{50} and GI_{50} values and presented the higher contents of total phenolic compounds in comparison to other fractions. The phenolic compounds profile of the fractions was determined. The bio-guided fractionation of the extracts resulted in several fractions with improved bioactivity relative to the corresponding extracts. Their lower compositional complexity allowed the identification of more than two dozen compounds, to the best of our knowledge, so far not reported in G. robertianum.

1. Introduction

Geranium robertianum L., commonly known as Herb Robert or Red Robin, belongs to Geraniaceae family and is commonly found in woodlands, waste lands, roadsides, hedge banks or old walls ([Cunha](#page-9-0) [et al., 2012](#page-9-0)). Widely spread across Europe, it can also found in Asia, Africa and America [\(Allen and Hat](#page-8-0)field, 2004). This plant has been used in folk medicine and herbalism practice for the treatment of a variety of digestive system disorders, and also as anti-inflammatory, haemostatic, antidiabetic, antibacterial, antiallergic, anti-cancer and diuretic [\(Graça](#page-9-1) [et al., 2016a](#page-9-1)).

As a consequence of the lifestyle, the contemporary society is subjected to continuous exposure to various sources of aggressors which can provoke an increase in the production of reactive oxygen species (ROS). Free radicals have been implicated in the onset of various diseases such as cardiovascular diseases, cancer, cataracts, age-related decline in the immune system, and degenerative diseases. Antioxidants have the ability to neutralize these species, protecting cells ([Aprioku,](#page-8-1) [2013;](#page-8-1) [Carocho and Ferreira, 2013;](#page-8-2) [Rubió et al., 2013\)](#page-9-2).

G. robertianum contains many biologically active compounds, mostly polyphenols [\(Amaral et al., 2009; Fodorea et al., 2005;](#page-8-3) [Ivancheva and Petrova, 2000; Kartnig and Bucar-Stachel, 1991;](#page-8-3)

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[Kobakhidze and Alaniya, 2004; Neagu et al., 2013; Paun et al., 2011,](#page-8-3) [2012, 2014](#page-8-3)).

The antioxidant effect of phenolic compounds is mainly due to their redox properties and their capacity to block the production of ROS such as superoxide, hydrogen peroxide and hydroxyl, preventing, for example, DNA oxidative damage and/or lipid peroxidation [\(Rubió et al.,](#page-9-2) [2013\)](#page-9-2). Flavonoids are the main phenolic compounds found in G. robertianum ([Graça et al., 2016a](#page-9-1)), and their effects on human health are very often ascribed to their potential ability to act by diminishing free radical steady-state concentration in biological systems and so providing antioxidant protection [\(Galleano et al., 2010\)](#page-9-3).

The antioxidant activity and the cytotoxicity against several human tumor cell lines and non-tumor porcine liver primary cells of several aqueous and organic extracts of Geranium robertianum L. were recently evaluated by our research group ([Graça et al., 2016b](#page-9-4)). In the present work, the more active extracts − ethyl acetate, acetone and methanol − were fractionated by gradient elution column chromatography on silica gel and the resulting fractions were evaluated for their bioactive properties, namely the antioxidant and antiproliferative properties. The profile of phenolic compounds of the different fractions was determined and correlated with their bioactivity.

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2. Materials and methods

2.1. Sample

Geranium robertianum L. was collected in França, Serra de Montesinho, Bragança, Northeastern Portugal, in May 2015. The botanical identification was confirmed by Ana Maria Carvalho from the Department of Biology and Biotechnology of the School of Agriculture, Polytechnic Institute of Bragança (Trás-os-Montes, Portugal). Voucher specimens (ETBO 62 May 2015) are deposited at the herbarium of the Escola Superior Agrária de Bragança (BRESA). The ethyl acetate, acetone and methanol extracts were obtained by sequential extraction of the whole plant, as described previously ([Graça et al., 2016b](#page-9-4)).

2.2. Standards and reagents

Silica gel 0.060–0.200 mm, 60 A was obtained from Acros Organics (Geel, Belgium). Acetonitrile was of HPLC grade from Fisher Scientific (Lisbon, Portugal). Trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2 carboxylic acid), acetic acid, formic acid, ellipticine, sulphorhodamine B (SRB), trypan blue, trichloroacetic acid (TCA) and Tris was purchased from Sigma-Aldrich (St. Louis, MO, USA). 2,2-Diphenyl-1-picrylhydrazyl (DPPH) was obtained from Alfa Aesar (Ward Hill, MA, USA). Phenolic compound standards were purchased from Extrasynthèse (Genay, France). Foetal bovine serum (FBS), L-glutamine, Hank's balanced salt solution (HBSS), trypsin-EDTA (ethylenediaminetetraacetic acid), penicillin/streptomycin solution (100 U/mL and 100 mg/mL, respectively), RPMI-1640 and DMEM media were from Hyclone (Logan, UT, USA). Water was treated in a Milli-Q water purification system (TGI Pure Water Systems, Greenville, SC, USA).

2.3. Fractionation of extracts

2.3.1. Ethyl acetate extract

The extract (0.7 g) was dissolved in a minimum of CH_2Cl_2 , mixed with silica gel, and the mixture evaporated to dryness and placed on the top of silica gel column. The dry-loaded extract was fractionated by gradient elution column chromatography using: CH_2Cl_2 ; $CH_2Cl_2/EtOAc$ − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); EtOAc, EtOAc/acetone − (90:10), (80:20), (70:30), (60:40), (50:50), (60:40), (70:30), (80:20), (90:10); acetone; acetone/MeOH − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); MeOH; MeOH/formic acid (99:1), (97:3), (95:5). Three hundred and eighty two samples (∼23 mL each) were collected and grouped in eleven fractions (FEA1-11) according to the similarity of their TLC profiles. The solvent of these final fractions was removed under reduced pressure at 40 °C.

2.3.2. Acetone extract

The extract (3 gr) was dissolved in acetone, mixed with silica gel, and the mixture evaporated to dryness and placed on the top of a silica gel column. The dry-loaded extract was fractionated by gradient elution column chromatography using: CH₃Cl; CH₃Cl/acetone − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); acetone; acetone/MeOH − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:7orated to dryness and placed on the top of a silica gel column. The dry-loaded extract was fractionated by gradient elution column chromatography using: CH₃Cl; CH₃Cl/acetone – (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); acetone; acetone/MeOH − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:7orated to dryness and placed on the top of a silica gel column. The dry-loaded extract was fractionated by gradient elution column chromatography using: CH_3Cl ; $CH_3Cl/acetone - (90:10)$, (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); acetone; acetone/MeOH − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); MeOH; MeOH/formic acid (99:1), (97:3), (95:5). Six hundred and seventy-three fractions (∼23 mL each) were collected and combined on the basis of their TLC profiles to yield eleven final fractions (FA1-11) which were evaporated to dryness under reduced pressure at 40 °C.

2.3.3. Methanol extract

The extract (4 gr) was dissolved in MeOH, mixed with silica gel, and the mixture evaporated to dryness and placed on the top of a silica gel column. The dry-loaded extract was fractionated by gradient elution column chromatography using: CH_2Cl_2 ; $CH_2Cl_2/EtOAC - (80:20)$, (40:60), (20:80); EtOAc, EtOAc/acetone − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (20:80), (10:90); acetone; acetone/MeOH − (90:10), (80:20), (70:30), (60:40), (50:50), (40:60), (30:70), (20:80), (10:90); MeOH; MeOH/formic acid (99:1), (97:3), (95:5). Three hundred and forty three fractions (∼23 mL each) were obtained, checked by TLC and those with similar band patterns were combined, yielding thirteen final fractions (FM1-13). The solvent was removed at 40 °C under reduced pressure.

2.4. Phenolic profile of the fractions

The fractions were re-dissolved in water/methanol 80:20 (v/v) (final concentration 5 mg/mL). Phenolic compounds were determined by high performance liquid chromatography with a diode array detector, coupled to mass spectrometry using the electrospray ionization interface (HPLC-DAD-ESI/MS), as previously described [\(Dias et al.,](#page-9-5) [2013; Roriz et al., 2014](#page-9-5)). Double online detection was carried out with a DAD using 280 nm and 370 nm as preferred wavelengths and with a mass spectrometer (MS) connected to the HPLC system via the DAD cell outlet. The phenolic compounds were identified by comparing their retention time, UV–vis and mass spectra with those obtained from standard compounds, when available. Otherwise, peaks were tentatively identified comparing the obtained information with the available data reported in the literature. For quantitative analysis, a calibration curve for each available phenolic standard $(2 - 100 \mu g/mL)$ was constructed based on the UV signal. The available phenolic standards are: chlorogenic acid $(y = -161172 + 168823x, r^2 = 0.9999)$; protocatechuic acid ($y = 27102 + 214168x$, $r^2 = 0.9999$); ferulic acid $(y = -185462 + 633126x,$ $r^2 = 0.999$; p-hydroxibenzoic acid $(y = 173056 + 208604x, r² = 0.9995)$; vannilic acid $(y = -28661 +$ 29751x, $r^2 = 0.9999$; caffeic acid (y = 406369 + 388345x, $r^2 = 0.9939$; quercetin-3-O-glucoside (y = -160173 + 34843x, $r^2 = 0.9998$); ellagic acid (y = -317255 + 26719x, $r^2 = 0.9986$); p coumaric acid (y = 6966.7 + 301950x, r^2 = 0.9999); quercetin-3-0rutinoside (y = 76751 + 13343x, r^2 = 0.9998); kaempferol-3-O-rutinoside (y = 30861 + 11117x, $r^2 = 0.9999$); naringenin (y = 78903 + 18433x, $r^2 = 0.9998$). For the identified phenolic compounds for which a commercial standard was not available, the quantification was performed through the calibration curve of other compound from the same phenolic group. The results are expressed in mg per g of extract.

2.5. Evaluation of bioactivity of the fractions

2.5.1. Antioxidant activity

Three different in vitro assays were performed using solutions prepared by serial dilution of the stock solutions: scavenging of DPPH (2,2 diphenyl-1-picrylhydrazyl) radicals, reducing power (measured by ferricyanide Prussian blue assay) and inhibition of β-carotene bleaching as previously described [\(Dias et al., 2013; Roriz et al., 2014\)](#page-9-5)

2.5.2. Cytotoxicity in human tumor cell lines and hepatotoxicity in nontumor cells

Four human tumor cell lines were tested using solutions prepared by serial dilution of the stock solutions: MCF-7 (breast adenocarcinoma), NCI-H460 (non-small cell lung cancer), HeLa (cervical carcinoma) and HepG2 (hepatocellular carcinoma), as previously described [\(Dias et al.,](#page-9-5) [2013\)](#page-9-5). The hepatotoxicity was evaluated against a non-tumor porcine liver primary cells (PLP2), as described earlier ([Barros et al., 2013; Dias](#page-8-4) [et al., 2013](#page-8-4)).

2.6. Statistical analysis

For all the experiments, three samples were analysed and all the assays were carried out in triplicate. The results are expressed as mean values \pm standard deviation (SD). All statistical tests were performed at a 5% significance level using IBM SPSS Statistics for Windows, version 23.0. (IBM Corp., Armonk, NY, USA). The fulfilment of the oneway ANOVA requirements, specifically the normal distribution of the residuals (data not shown) and the homogeneity of variance, was tested by means of the Shapiro Wilk's and the Levene's tests, respectively. All dependent variables were compared using Tukey's honestly significant difference (HSD) or Tamhane's T2 multiple comparison tests, when homoscedasticity was verified or not, respectively. Furthermore, a Pearson's correlation analysis between the antioxidant activity and all the analysed compounds was carried out, with a 95% confidence level.

3. Results and discussion

3.1. Analysis of phenolic profile of the fractions

Peak characteristics and tentative identities are presented in [Table 1](#page-3-0) and the quantification of each individual compound present in each fraction is presented in [Tables 2](#page-4-0)–4.

The most abundant compounds were assigned to hydrolyzable tannins, being compounds 2, 9, 12, 17, 21, 23, 27 and 35 identified as gallotannins composed by two, three, four or five galloyl moieties linked to glucose. The mass spectra characteristics of these compounds consisted of the deprotonated molecule ($[M - H]$ ⁻ ions at m/z 483 (compound 2), 635 (compounds 9, 12 and 17), 787 (compounds 21, 23 and 27), and 939 (compound 35), with the loss of one or more galloyl groups (152 u) and/or gallic acid (170 u). Compounds 14, 15, 18 ([M $-$ H]⁻ ion at m/z 951) and 30 ([M − H]⁻ ion at m/z 933), which were abundantly present in the G. robertianum acetone and MeOH extracts, were tentatively identified as ellagitannins: three isomers of geraniin and castalagin/vescalagin (both with the same molecular weight), respectively. The different chromatographic retention times of compounds 14, 15 and 18, presenting the same molecular mass, strongly suggests that they correspond to distinct isomers. These compounds have also been reported in different Geranium species ([Okuda et al.,](#page-9-6) [2000; Tuominen, 2013; Tuominen et al., 2013\)](#page-9-6) and previously reported by us in G. robertianum [\(Graça et al., 2016b\)](#page-9-4). Compound 8 ([M−H][−] at m/z 633) was identified as galloyl-HHDP-glucose, while peaks 19 and 26 ([M−H][−] at m/z 785) were coherent with two digalloyl-HHDPglucose isomers. The pseudomolecular ion $[M - H]$ ⁻ of compound 32 (m/z at 935) might point to a galloyl-bis-HHDP-glucose, although a different fragmentation pattern was described ([Dias et al., 2015;](#page-9-7) [Hanhineva et al., 2008\)](#page-9-7). However, the presence of two HHDP moieties in its structure was supported by the loss of an HHDP fragment (302 Da, from the transition 935 > 633), together with the observation of a product ion at m/z 301. All of these mentioned compounds have already been described in the previously study G. robertianum [\(Graça](#page-9-4) [et al., 2016b\)](#page-9-4).

Peaks 6 and 38 were tentatively associated to ellagitannins, based on their UV spectra and on the observation of an $MS²$ fragment ion at m/z 301 ([HHDP-H]⁻). Nevertheless, a full identification of these compounds was not possible being identified as unknown ellagitannins.

Seventeen flavonoid glycosides derivatives were detected nine of which were kaempferol derivatives, seven quercetin derivatives and a narigenin derivative. Peaks 25 (quercetin-3-O-rutinoside), 34 (kaempferol-3-O-rutinoside), 36 (quercetin-3-O-glucoside) and 39 (narigenin-7-O-glucoside), were positively identified according to their retention

time and UV and mass spectra, in comparison with commercial standards.

Compounds 28, 29, 33, 43, 47, 48, 49 and 50 were identified as kaempferol glycoside derivatives owing to the product ion observed at m/z 285 and UV spectra (λ_{max} around 348 nm). Compound 28 ([M-H]⁻ at m/z 765) presented two $MS²$ fragments, revealing the alternative loss of acetylglucuronyl (m/z at 547; -42–176 u) and malonylglucuronyl (m/z at 285; −86–176 u) residues, indicating location of each residue on different positions of the aglycone. For compound 29 ([M-H]⁻ at m/z 739) only one $MS²$ fragment was present (m/z) at 285; -146-146-162 u) suggesting that the three sugars were linked together. No information about the identity of the sugar moieties and location onto the aglycone could be obtained, so, these compounds were tentatively identified as kaempferol-O-acetylglucuronyl-Omalonylglucuronide and kaempferol-O-deoxyhexosyl-(deoxyhexosyl-hexoside), respectively. Mass characteristics of peak 47 ($[M-H]$ ⁻ at m/z 609), 43 and 49 ($[M-H]$ ⁻ at m/z 637) indicated that these compounds corresponds to kaempferol derivatives bearing two hexosyl (-162–162 u) and glucuronyl (-176–176 u) residues, respectively. The observation of only one $MS²$ fragment at m/z 285, also indicated that the residue moieties were linked together. Thus, these compounds were tentatively identified as kaempferol-Odihexoside and kaempferol-O-diglucuronide. Similarly, compounds 48 ([M-H][−] at m/z 607) and 50 ([M-H][−] at m/z 591), also released one fragment at m/z 285, from the respective losses of deoxyhexosyl-glucuronide (322 u) and acetylpentosyl-pentoside (306 u) moieties, thus being tentatively assigned as kaempferol-O-deoxyhexosyl-glucuronide and kaempferol-O-acetylpentosylpentoside. Compound 33 presented the same molecular weight as compound 34 (kaempferol-3-O-rutinoside), but with an earlier retention time, thus being assigned to kaempferol-O-deoxyhexosyl-hexoside.

Compounds 20, 24, 31, 37 and 41 were identified as quercetin glycosides derivatives based on their UV spectra (λ_{max} around 350 nm) and the production of a MS^2 fragment ion at m/z 301. Tentative identities of these compounds were assigned based on their pseudomolecular ions using a similar reasoning as for kaempferol derivatives. Thus, compound 20 ([M-H]⁻ at m/z 755), 24 and 31 ([M-H]⁻ at m/z 609) could correspond to quercetin-O-dideoxyhexosyl-hexoside and quercetin-O-deoxyhexosyl-hexoside, respectively, whereas peaks 37 ([M-H][−] at m/z 597) and 41 ([M-H][−] at m/z 623) should correspond to quercetin-O-deoxyhexosyl-pentoside and quercetin-O-deoxyhexosylglucuronide, respectively.

Peaks 20, 25, 29, 33, 34 and 36, correspond to different quercetin and kaempferol glycosides that have been previously described in G. robertianum [\(Fodorea et al., 2005](#page-9-8); [Graça et al., 2016b;](#page-9-4) [Ivancheva and](#page-9-9) [Petrova, 2000](#page-9-9); [Kartnig and Bucar-Stachel, 1991](#page-9-10); [Kobakhidze and](#page-9-11) [Alaniya, 2004](#page-9-11)).

Twelve phenolic acids were detected, five hydroxybenzoic acids (compounds 3, 11, 13, 40 and 42) and seven hydroxycinnamic acid and derivatives (compounds 1, 4, 5, 7, 10, 16 and 22). Protocatechuic acid (compound 3), 5-O-caffeoylquinic acid (compound 5), p-hydroxybenzoic acid (compound 11), vanillic acid (compound 13), p-coumaric acid (compound 22) were positively identified according to their retention time and UV and mass spectra, in comparison with commercial standards. 3-O-Caffeoylquinic acid (compound 1), 3-p-coumaroylquinic acid (compound 4), 3-O- and 4-O-feruloylquinic acids (compounds 7 and 10, respectively), were tentatively identified taking into account their fragmentation pattern and relative intensities similar to those reported by Cliff[ord et al. \(2003\)](#page-8-5) and Cliff[ord et al. \(2005\)](#page-9-12). Compound 16 ($[M-H]$ ⁻ at m/z 335) was tentatively identified as a caffeoylshikimic acid due to its similar fragmentation pattern described by [Al-Ayed \(2015\)](#page-8-6) and [Bastos et al. \(2007\).](#page-8-7) Compounds 40 ([M-H][−] at m/z 625) and 42 ([M-H]⁻ at m/z 591), both produced a major MS² fragment ion at m/z 301, which can be interpreted as corresponding to deprotonated ellagic acid; in the case of compound 40, it lost two hexosyl moieties (–324 u), which was tentatively identified as an ellagic acid dihexoside. No full identity was obtained for compound 42, being assigned as ellagic acid derivative. Moreover, no tentative identification was possible to achieve for compounds 44, 45 and 46, remaining as

Retention time (Rt), wavelengths of maximum absorption in the visible region (λ_{max}), mass spectral data, tentative identification of phenolic compounds of the extracts of Geranium robertianum L.

d − detected. nd − not detected.

^a Hitherto unreported in G. robertianum.

unknown compounds. 5-O-Caffeoylquinic acid, 3-O-caffeoylquinic acid, p-coumaric acid have been reported in some Geranium species [Paun](#page-9-13) [et al., 2011, 2012; Tuominen, 2013; Tuominen et al., 2013\)](#page-9-13).

In this work, the lower compositional complexity of the fractions obtained compared with the whole extracts turned possible the identification of more than two dozens of compounds, mostly quercetin and kampferol glycosides, phenolic acids and galloyl tannins [\(Table 1](#page-3-0)), which, to the best of our knowlege, have not hitherto been reported in G. robertianum.

3.2. Evaluation of bioactivity of the fractions

The antioxidant properties of the fractions obtained from the most bioactive extracts of G. robertianum (EtOAc, acetone and MeOH) were evaluated by three different assays: DPPH radical scavenging capacity, reducing power and β-carotene bleaching inhibition. The results are shown in [Table 5.](#page-7-0) The cytotoxic activities of the same fractions were tested against four human tumor cell lines (i.e., MCF-7, NCI-H460, HeLa and HepG2) and in a non-tumor primary culture (porcine liver primary cell culture). The results are presented in [Table 5.](#page-7-0)

Qu[a](#page-4-1)ntification of phenolic compounds present in the fractions of ethyl acetate extract of Geranium robertianum L.^a.

TPA − Total phenolic acids. TED − Total ellagitannins derivate. TF − Total flavonoids. TPC − Total phenolic compounds. FEA − fractions of ethyl acetate extract. ^a Fractions FEA1, FEA2, FEA3 and FEA4 did not present phenolic compounds.

TPA - Total phenolic acids. TED - Total ellagitannins derivate. TF - Total flavonoids. TPC - Total phenolic compounds. FA - fractions of acetone extract. ^a Fractions FA1, FA2 and FA3 did not present phenolic compounds.

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Table 4 (continued)

Table 4 (continued)

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In general all fractions resulting from the acetone extract (FA1- FA11) exhibited bioactivity. The most bioactive ones were FA4, FA5 and FA8, which also presented the higher contents of total phenolic compounds (91.0, 97.3 and 48.1 mg/g, respectively, [Table 3](#page-4-3)) when compared to the other fractions. Amongst those, fraction FA4 displayed particularly low hepatotoxicity ($IC_{50} > 400 \mu g/mL$).

Fractions FA4 and FA5 showed to contain only ellagitannins and the higher amount of these group of compounds among all the tested fractions. Compound 30, which was tentatively identi fied as castalagin or vescalagin, was the main compound of these fractions (23.6 mg/g in FA4 and 31.5 mg/g in FA5, [Table 3](#page-4-3)). The antioxidant and cytotoxic properties of both castalagin and vescalagin have been previously described ([Fernandes et al., 2009](#page-9-14); [Fridrich et al., 2008](#page-9-15); [Jordão et al.,](#page-9-16) [2012\)](#page-9-16). In each of the fractions FA4, FA5 and FA8 compound 14, an isomer of geraniin, was the second most abundant compound present (9.3, 26.6 and 15.1 mg/g, respectively, [Table 3\)](#page-4-3). Geraniin has been described as the main phenolic compound in various Geranium species ([Harborne and Williams, 2002; Okuda et al., 1980, 1979, 2000;](#page-9-17) [Piwowarski et al., 2014; Tuominen, 2013; Tuominen et al., 2013; Wu](#page-9-17) [et al., 2011](#page-9-17)) and is endowed with several bene ficial biological activities ([Cheng et al., 2017](#page-8-8)). Apart from the identi fied ellagitannins, FA8 showed to contain also some phenolic acids and flavonoids. In this fraction, compound 32, tentatively identi fied as galloyl-bis-HHDP-glucose, was the main compound present (15.5 mg/g, [Table 3\)](#page-4-3).

The MeOH extract fractionation resulted in thirteen fractions (FM1- 13). In general, all of them possessed biological activity. FM3 and FM4 were the fractions that exhibited, simultaneously, the best antioxidant and cytotoxic activities. FM2 also exhibited good cytotoxic activity but poor/moderate antioxidant activity, except for the DPPH assay. FM4 showed additionally the highest content of total phenolic compounds (63.9 mg/g, [Table 4\)](#page-5-0), but this direct relation activity/content of phenolic compounds was not observed for the remaining more bioactive fractions.

Fraction FM2 possessed approximate amounts of ellagitannins and flavonoids (11.43 and 9.30 mg/g, respectively, [Table 4](#page-5-0)), and also some phenolic acids (2.23 mg/g, [Table 4](#page-5-0)). Compound 48, which was tentatively identified as kaempferol-O-deoxyhexosyl-glucuronide, was the main compound detected (4.78 mg/g, [Table 4](#page-5-0)). Several biological properties, including antioxidant and antiproliferative activities, of di fferent kaempferol glycosides has been previously reported ([Calderón-Montaño et al., 2011](#page-8-9) ; [Xiao et al., 2014\)](#page-9-18). The second most abundant compound in this fraction was tetragalloyl-glucose (compound 27) The biological activity of this compound has been described ([Sugimoto et al., 2009; Xiang et al., 2010; Zhang et al., 2009](#page-9-19)). FM3 showed to be constituted by phenolic acids, flavonoids and ellagitannin derivatives, this later group being the most abundant (13.1 mg/g, [Table 4\)](#page-5-0). Fraction FM4 consisted mainly of ellagitannin derivatives (57.8 mg/g, [Table 4\)](#page-5-0), together with a small amount of flavonoids (6.06 mg/g, [Table 4\)](#page-5-0). Compound 14 (geraniin isomer 1) was the most abundant in both fractions FM3 and FM4 (5.1 and 38.7 mg/g, respectively, [Table 4](#page-5-0)).

Fractions FM7, FM8, FM11 and FM12 displayed high antioxidant activity but no relevant cytotoxic activity. With the exception of FM7, for which phenolic acids were the major constituents, in all the other fractions ellagitannin derivatives were the most abundant compounds. The major compound found in FM7 was tentatively identified as 3-Oca ffeoylquiic acid (compound 1), while for the remaining fractions geraniin isomer 1 (compound 14) was the most abundant compound ([Table 4](#page-5-0)).

The EtOAc extract yielded, after fractionation, eleven fractions (FEA1-11). However, the color of fractions FEA1-FEA4 did not allow the study of their bioactivity, most probably due to the presence of chlorophylls or other pigments. The remaining fractions of this extract possessed antioxidant activity much lower than that of the fractions resulting from the acetone and MeOH extracts. Fractions FEA5, FEA6 and FEA8 showed moderate cytotoxicity. Fraction FEA5 showed to

Bioactive properties of different fractions of ethyl acetate, acetone and methanol extracts of Geranium robertianum L.

FEA − fractions of ethyl acetate extract; FA − fractions of acetone extract; FM − fractions of methanol extract; nd − not determined. The antioxidant activity was expressed as EC₅₀ values, what means that higher values correspond to lower reducing power or antioxidant potential. EC₅₀: Extract concentration corresponding to 50% of antioxidant activity or 0.5 of absorbance in reducing power assay. Trolox EC₅₀ values: 41 μg/mL (reducing power), 42 μg/mL (DPPH scavenging activity) and 18 μg/mL (β-carotene bleaching inhibition). GI₅₀ values correspond to the sample concentration achieving 50% of growth inhibition in human tumor cell lines or in liver primary culture PLP2. Ellipticine GI₅₀ values: 1.21 μg/mL (MCF-7), 1.03 μg/mL (NCI-H460), 0.91 μg/mL (HeLa), 1.10 μg/mL (HepG2) and 2.29 μg/mL (PLP2).

¹ Homoscedasticity among fractions of each exctract were tested by the Levene test: homoscedasticity, $p > 0.05$; heteroscedasticity, $p < 0.05$.

 $2 p < 0.05$ indicates that the mean value of at least one component differs from the others (in this case, multiple comparison tests were performed). For each fraction, means within a column with different letters differ significantly ($p < 0.05$).

contain only phenolic acids and, at the same time, the higher amount of this phenolic group among all the tested fractions. The major compound found in this fraction was p-hydroxybenzoic acid (compound 11, [Table 2\)](#page-4-0). Fractions FEA6 and FEA8 showed to contain, apart from phenolic acids, flavonoids as the main phenolic compounds. In both fractions, the most abundant compounds were derivatives of quercetin and kaempferol ([Table 2](#page-4-0)).

The bioactivity of several fractions was found to be higher, in some cases considerably, than that of the corresponding extract. However, for the majority of fractions the bioactivity was lower what seems to point to a synergistic effect of the mixture of compounds present in the extract. In a number of fractions the hepatotoxicity displayed $GI₅₀$ values higher than 400 mg/mL, while for the whole extracts $GI₅₀$ values ranged from 176 to 290 mg/mL.

3.3. Correlation between the phenolic profile and the bioactivity of the fractions

The relationship between the sample's cytotoxic and antioxidant properties and the phenolic composition was established based on the correlation factors between total phenolic acids, total ellagitannin derivatives, total flavonoids and total phenolic compounds, and the EC_{50} and $GI₅₀$ values found, respectively, for the antioxidant and cytotoxic activities [\(Table 6](#page-8-10)). In what concerns the fractions obtained from the EtOAc extract, there seems to exist no relevant correlation between phenolic composition and antioxidant activity. However, the cytotoxic activity of these fractions was correlated significantly with total flavonoids (r^2 = −0.814, −0.834 and −0.739, for the NCI-H460, HepG2 and PLP2 cell lines, respectively). The total phenolic compounds of

Correlation coefficients of phenolic compounds with bioactivity of the fractions of Geranium robertianum L. extracts.

TPA − Total phenolic acids. TED − Total ellagitannins derivate. TF − Total flavonoids. TPC − Total phenolic compounds. r ² − Correlation coefficient.

these fractions also showed significant correlation ($r^2 = -0.863$) with the toxicity against the HepG2 cell line.

Acetone fractions showed the highest correlation between total phenolic acids and the antioxidant activity determined by the β-carotene/linoleic acid assay ($r^2 = -0.959$), and a significant correlation $(r^2 = -0.784)$ between the total phenolic acids and the antioxidant activity assessed by the reducing power assay. The total ellagitannin derivatives showed good correlation with the results of the DPPH assay $(r^2 = -0.778)$. The total phenolic compounds and the antioxidant activity of the acetone fractions showed to be significantly correlated $(r² = -0.710, -0.842$ and −0.801, for DPPH, reducing power and βcarotene/linoleic acid assays, respectively). The cytotoxic effects of these fractions and the respective phenolic composition also presented relatively good correlation factors, being the cytotoxicity displayed against the NCI-H460 cell line correlated with total phenolic acids $(r^2 = -0.704)$ and the cytotoxicity observed against the NCI-H460, HeLa and HepG2 cell lines correlated with total phenolic compounds $(r^2 = -0.853, -0.755$ and -0.698 , respectively). The activities of the fractions obtained from the MeOH extract revealed to be poorly correlated with the phenolic composition. The best correlation was found between total phenolic compounds and the cytotoxicity observed on the HepG2 cell line $(r^2 = -0.69)$.

4. Conclusions

Following a recent work conducted by our research group in which the bioactivity of different organic extracts of G. robertianum was studied, in this work the most bioactive extracts (EtOAC, acetone and MeOH) were fractionated by gradient elution column chromatography. The phenolic profile of the resulting fractions was determinated and the antioxidant and cytotoxic activities were evaluated and correlated with the respective phenolic composition. In addition to hydrolyzable tannins, the most abundant compounds found in this plant, were also identified flavonoid glycosides derivatives − mainly kaempferol and quercetin derivatives – and phenolic acids. To the best of our knowledge, about two dozen of the compounds found were identified in G. robertianum for the first time. In general, the fractions from the acetone extract were those exhibiting higher bioactivity when compared with those from the other extracts. Three fractions of acetone extract (FA4, FA5 and FA8) displayed consistently low EC_{50} and GI_{50} values. These fractions also presented the higher contents of total phenolic compounds when compared with the other fractions. Fraction FA4

displayed particularly low hepatotoxicity $(IC_{50} > 400 \text{ µg/mL})$. Generally, the fractions resulting from the MeOH extract possessed moderate biological activity. The EtOAc fractions presented antioxidant activity much lower than that of the fractions resulting from the acetone and MeOH extracts. Some of these fractions showed moderate cytotoxicity. The acetone fractions were the ones that presented the highest correlation factors between biological activity and phenolic composition.

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