

Layer-by-layer deposition of antimicrobial polymers on cellulosic fibers: a new strategy to develop bioactive textiles

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In recent years, there has been an increase of infectious diseases caused by different microorganisms and the development of antibiotic resistance. In this way, the search for new and efficient antibacterial materials is imperative. The main polysaccharides currently used in the biomedical and pharmaceutical domains are chitin and its derivative chitosan (CH) and alginates (ALG). In this study, a simple technique of Layer by Layer (LbL) of applying polycation CH and polyanion ALG was used to prepare CH/ALG multilayers on cotton samples via the electrostatic assembly with success. The CH/ALG cotton samples (functionalized) were investigated for their antibacterial properties towards *Staphylococcus aureus* and *Klebsiella pneumoniae* using the international standard method JIS L 1902:2002. The antibacterial activity of the functionalized samples was tested in terms of bacteriostatic and bactericidal activity, and results showed that the samples exhibited a bacteriostatic effect on the two bacteria tested, as expected. In addition, samples with five layers (CH/ALG/CH/ALG/CH) were more effective in inhibiting bacterial growth. This new coating for cellulosic fibers is a new strategy and may open new avenues for the development of antimicrobial polymers with potential application in health-care field. Copyright © 2013 John Wiley & Sons, Ltd.

Keywords: chitosan; alginate; cotton; LbL; JIS L 1902

INTRODUCTION

The number of different antimicrobial agents suitable for textile application on the market has increased drastically in recent years. Several different types of antimicrobial agents, such as oxidizing agents, coagulants, diphenyl ether (bis-phenyl) derivatives, heavy metals and metallic compounds and quaternary ammonium compounds are used in the textile industry to confer antimicrobial properties.^[1] The selection of the antimicrobial agent depends on the mechanism of antimicrobial activity (bacteria and fungi), toxicity, application method and cost.^[1] Nowadays, the textile manufactures prefer eco-friendly chemicals to impart antimicrobial finishing on textiles.

Polysaccharide biopolymers including ALG and CH have been the focus of an expanding number of studies reporting their potential use in biomedical research applications such as cell encapsulation, drug delivery and tissue engineering. Therefore, multilayered films containing both biopolymers could be useful in the coating of substrates for different biomedical applications, in particular in the development of new wound dressings.^[2,3]

CH, a natural biopolymer, has a combination of many unique properties such as biodegradability, non-toxicity, cationic nature, antitumor, immunostimulatory^[4] and antimicrobial activity.^[5,6] CH is also known in the wound management field for its hemostatic properties.^[7]

ALG is a natural biopolymer, and it is non-toxic, biocompatible, biodegradable, less expensive and freely available.^[8,9] ALG is known to exhibit minimum cytotoxic effects and reduced hemolysis when in contact with blood.^[8]

Electrostatic LbL assembly is a versatile method of fabricating multilayer films and coating from materials in solution, notably, oppositely charged polyelectrolytes in solution. Polyelectrolytes

are polymers with ionizable groups along the chain, classified into anionic and cationic according to their functional group. An advantage of LbL is that the film is fabricated directly on the surface of interest. The method is based on the successive deposition of oppositely charged polymers onto solid surfaces.^[10,11] A deposition cycle creates a bilayer, and these cycles can be repeated as often as needed according to previous work.^[12] The number of deposition cycles and the types of polyelectrolytes used in the construction allow for the control of thickness and roughness of the multilayered film.^[13] Figure 1 can illustrate this method.

Numerous studies involving different polymer substrates and several synthetic polyelectrolytes have been published. But,

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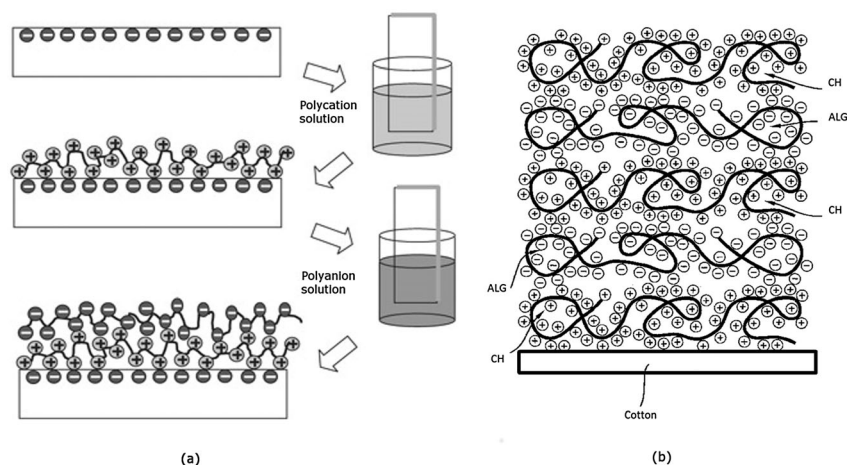


Figure 1. (a) The sequence of layer-by-layer electrostatic in negatively charged substrate, dipping into polycationic solution (CH), polycation layer deposited, dipping into polyanion solution (ALG), and the polyanion layer deposited. (b) Design of structure of the sample functionalized, adapted from Nabok, 2005.^[14]

there are very few studies concerning the deposition of alternate polyelectrolyte on natural textile supports. Hyde et al. (2005) reported a pioneering application of self-assembly technique on cotton substrates by LbL deposition of poly (styrene sulfonate) (PSS) and poly (allylamine hydrochloride) on cotton fibers.^[15] Dubas et al. (2006) have immobilized antimicrobial silver nanoparticles on nylon and silk fibers by LbL deposition method.^[16]

More recently, lamphaojeen et al. (2012), in this study, immobilized ZnO nanoparticles on cotton fabrics using PSS, through the LbL technique. As a result, they obtained a high UV protection factor and inhibition of the growth of *Staphylococcus aureus*.^[17]

Based on literature review, the aim of this study was to design a new process for the bio-functionalization of cellulosic fibers with an antimicrobial effect with potential application as a new substrate for wound dressings. For this purpose, natural antimicrobial ingredients (CH and ALG) were deposited on cotton samples by LbL and investigated as a new potential antibacterial coating for cellulose-based textiles with a broad range of application in hospital, medical and hygienic products where antimicrobial activity is of the utmost importance

EXPERIMENTAL

LbL coating of cotton

To apply the LbL technique, anionic cotton was prepared by using 2,2,6,6-tetramethylpiperidiny-1-oxy free radical (TEMPO) and NaBr, in order to give higher surface charge for polyelectrolyte deposition, as described elsewhere.^[18] Cotton samples were charged negatively by immersing cotton samples in TEMPO solution under moderate stirring, followed by a rinse with deionized water, according with literature.^[19,20] Chitosan of low molecular weight and 80% degree of deacetylation (CH, 1 mg/ml), a natural antimicrobial polyelectrolyte, and Alginate sodium salt (ALG, 1 mg/ml) solutions were prepared by dissolving CH and ALG in 0.1 M CH₃COOH and 0.5 M NaCl solutions, respectively. The pH values were adjusted to 5 using 0.1 M HCl and 1 N NaOH solutions.

The pH was selected to 5, to be approximately intermediate between the pK_a of CH (6.3) and ALG, pK_a of 3.38 and 3.65, as

previously reported by other authors.^[12,21] ALG and CH are two oppositely charged natural polyelectrolyte materials and very sensitive toward changes in external factor such as pH. At pH 5, the carboxylate group of ALG mainly exists in the form of COO⁻, and the amino group of CH mainly exists in the form of NH₃⁺. In this case, the presence of both COO⁻ and NH₃⁺ along polymer backbone could enhance the electrostatic interaction of the network structure^[22] (see Fig. 2).

Then, CH and ALG polyelectrolyte multilayer films were deposited over cotton by the LbL assembly, where CH was used as polycation and ALG as polyanion (see Fig. 2). For each layer deposition, the cotton substrate was immersed into the corresponding solution at room temperature for 5 min, followed by rinsing with deionized water to remove the excess of polyelectrolyte. Since the cotton samples were charged negatively, the CH was deposited as the first layer. Cotton samples treated during 30 min in TEMPO (designated by CT) and samples then functionalized by LbL were prepared with five layers (CH/ALG/CH/ALG/CH), six layers (CH/ALG/CH/ALG/CH/ALG), nine layers (CH/ALG/CH/ALG/CH/ALG/CH/ALG/CH/ALG) and ten layers (CH/ALG/CH/ALG/CH/ALG/CH/ALG/CH/ALG/CH/ALG), designated, respectively, by

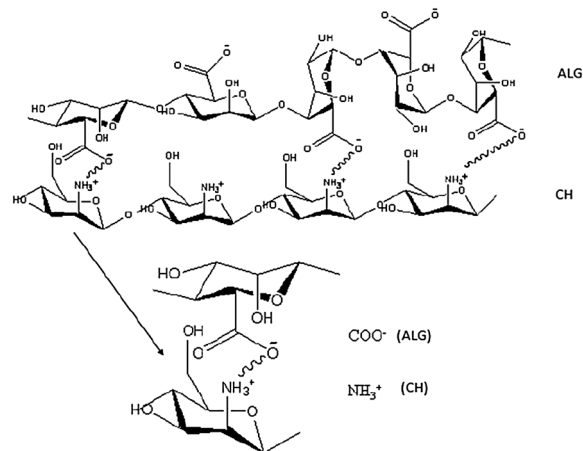


Figure 2. Schematic representation of the electrostatic interactions between the carboxylic groups of the ALG and the amine groups of the CH, adapted from Mi et al., 2002.^[23]

CT5, CT6, CT9 and CT10. After the last deposition, samples were dried in a desiccator at room temperature overnight. All chemicals were obtained from Sigma–Aldrich, are of analytical grade and were used as received.

Assessment of antibacterial activity – JIS L 1902:2002 - halo method

In order to evaluate the activity of antibacterial textiles, there are several standard methods available. The method used in this study and described in JIS L 1902 (Japanese Standard) appears to be the most commonly employed.

The culture medium Brain Heart Infusion used for the cultivation of the bacterial strains deployed in this work were prepared according with the instructions of the manufacturer. Culture media was dissolved directly after being weighed in deionized distilled water and then sterilized by autoclaving for 15 min at 121 °C. Agar was used to solidify the media before autoclaving. The strains of *S. aureus* and *Klebsiella pneumoniae* used were obtained from the American Type Culture Collection, ATCC 25923 and ATCC 13883, respectively. These strains were selected because they are indicated in several standards to evaluate antibacterial activity of textiles.^[24]

The cell density of bacterial suspensions was determined by measuring the optical density (OD) of appropriately diluted samples using a spectrophotometer at a wavelength of 640 nm [OD₆₄₀]. 100 µl aliquots of the appropriate dilutions were plated onto the surface of agar plates with a Mueller–Hinton medium, and after approximately 10 min, the functionalized cotton samples were placed on the top of agar plates and incubated for 24 h at 37 °C.^[25]

Assessment of antibacterial activity – JIS L 1902:2002 - absorption method

The JIS L 1902:2002 - absorption method is designed to quantitatively test the ability of textiles that have been treated with antibacterial agents to prevent bacterial growth and to kill bacteria, over an 18 h period of contact. This method is based on the quantitative determination of the potential effect and activity of functionalized samples, by the direct contact with a suspension of bacterial cells.

Textile samples with approximately 0.4 g, six control samples (without CH and ALG) and six samples functionalized with CH and ALG, previously sterilized in an autoclave at 121 °C for 15 min were tested for each bacterial strain. In order to calculate growth reduction rate, three samples were used to measure the number of live bacteria after inoculation (T_{0h}) and the other three to measure the number of live bacteria after incubation (T_{18h}).

Bacterial cell suspensions were collected from an overnight liquid culture in Nutrient Broth. After that, the bacterial concentration is adjusted to $1-2 \times 10^8$ cel/ml (equivalent to 0.5 McFarland), with the necessary dilutions to adjust the final bacterial concentration to $1 \pm 0.3 \times 10^5$ cel/ml.

Each textile sample was placed in a 50 ml Falcon tube, soaked with 200 µl of the inoculum previously prepared, and T_{18h} tubes were incubated for 18 h at 37 °C. For the release of bacterial cells from the textile samples, before and after the 18 h incubation period, 20 ml of 0.85% NaCl with surfactant Tween 80 (0.2%) was added to the samples in 50 ml Falcon tubes and vortexed. The resulting suspensions were used for the determination of

viable counts using serial dilutions prepared in sterile 0.85% sodium chloride solution and plated. The plates were incubated at 37 °C for 18 h, and the number of colonies was counted visually using a microscope. This procedure was performed in triplicate.^[25] The growth reduction rate of the bacteria was calculated using the equation:

$$\frac{T_{0h} - T_{18h}}{T_{0h}} \times 100\% = \text{reduction rate (\%)}$$

where, T_{0h} is the CFU/ml (CFU = colony forming units) of bacterial colonies at the initial stage (0h), and T_{18h} is the CFU/ml of bacterial colonies after 18 h incubation.^[24]

In order to carry out the judgement of test effectiveness, the growth value was calculated according to the following equation:

$$F = M_b - M_a$$

When the growth value is more than 1.5, the test is judged to be effective, and when the growth value is 1.5 or less, the test is judged to be not effective. When the test is not effective, a retest is necessary.

When the quantitative test has been effective, the bacteriostatic activity value should be calculated in accordance with the following equation:

$$S = M_b - M_c$$

and the bactericidal activity according to:

$$L = M_a - M_c$$

Where, F is the growth value, and S and L are the bacteriostatic and bactericidal activity values, respectively. M_a is the average of common logarithm of number of living bacteria of three test pieces immediately after inoculation of inoculum on standard cloth. M_b is the average of common logarithm of number of living bacteria of three test pieces after 18 h incubation on standard cloth. M_c is the average of common logarithm of number of living bacteria of three test pieces after 18 h incubation on antibacterial treated sample.^[25] Traditionally, bacteriostatic means prevention of multiplication of bacteria without destroying them, whereas bactericidal effect implies forthright killing of the organisms^[26]

RESULTS AND DISCUSSION

Assessment of antibacterial activity – JIS L 1902:2002 - halo method

The antibacterial efficacy of the CH/ALG multilayers was evaluated by assessing the reduction in bacterial growth on the functionalized cotton substrates. Antibacterial activity analysis revealed that on the CH/ALG functionalized cotton, the densities of attached bacteria decreased when compared with that of the control (cotton), as can be observed in Fig. 4 by the formation of a small halo on the functionalized samples. It appears that the effectiveness in preventing bacterial adhesion is high regardless of whether CH/ALG is deposited on cotton samples. This is in accordance with the expected. The use of CH as an anti-biofilm coating for medical applications has been suggested, as coating surfaces with CH is highly effective at restricting or preventing the formation *S. aureus* and *K. pneumoniae* biofilms.^[27]

Cationic polyelectrolytes, as well as other molecules with a net positive charge, are capable of killing microorganisms. The mechanism of antibacterial action of cationic polyelectrolytes is not completely understood, but it has been suggested that these polymers can interact electrostatically with anionic groups at the bacterial cell walls causing an increase of membrane permeability and subsequent leakage of cellular proteins which ultimately leads to cell death.^[28] In the context of this research, the observed antibacterial action of the polyelectrolytes is an interesting finding because in principle the antimicrobial activity of the functionalized samples can be adjusted using favorable layers number of polyelectrolytes on top of cotton.

On the other hand, according with several authors, the pH value affects the antibacterial effect, where an increase in pH leads to a decrease of the antibacterial action. In the case of CH, the high pH few amino groups in CH molecules will be free, but at pH below 6, the amino groups become ionized.^[4] In this study, we use pH 5; this implies a higher number of side amino groups available.

Likewise, the results of the JIS L 1902:2002 - qualitative method were used to assess the antibacterial activity of textile specimens.

In Fig. 3, a positive control (sample known to have antimicrobial activity) presents a halo around the sample. Halo size provides some indication of the potency of the antimicrobial activity of textile samples and also that the antimicrobial agent is released from the textile. In contrast, the results of the functionalized cotton (Fig. 4) showed an ambiguous inhibitory effect against *S. aureus* and *K. pneumoniae*, although in terms of the surrounding clearing zone, the different samples show the same inhibitory effect against all tested microorganisms.

Figure 4 (a) sample CT5, (b) samples CT6, CT9 and CT10, showed a small but clear inhibitory zone for *S. aureus*. Figure 4 (c) sample CT5, (d) samples CT6, CT9 and CT10, showed a smaller halo for the *K. pneumoniae*. However, Fig. 4 (a) samples CO (cotton) and CT, and Fig. 4 (c) samples CO and CT, showed a very thin inhibition line in the edges of the sample for both tested microorganisms. As a result, we conclude that for the *S. aureus* and *K. pneumoniae*, the control samples (CO and CT) showed little microbial growth inhibition. In contrast, the inhibition was more significant in the presence of CH/ALG in samples CT5, CT6, CT9 and CT10.

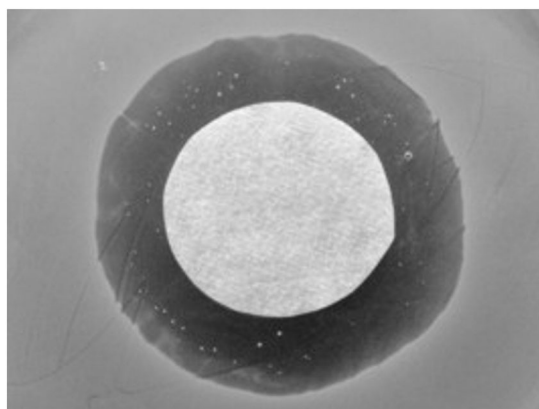


Figure 3. Image of a halo formation of a positive control (analysed using the standard method JIS L 1902:2002 - halo method, against *Staphylococcus aureus*).

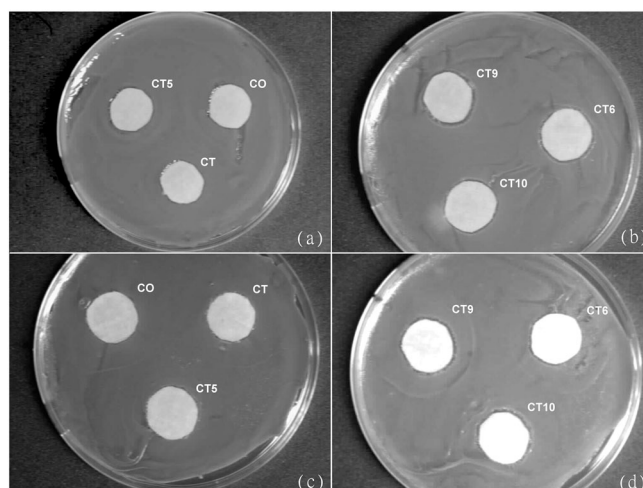


Figure 4. Images of samples tested according with the standard method JIS L 1902:2002 - halo method. a) and b), inhibition zone against *Staphylococcus aureus* for CO, CT, CT5, CT6, CT9 and CT10 samples. c) and d), inhibition zone against *Klebsiella pneumoniae* for CO, CT, CT5, CT6, CT9 and CT10 samples.

From the little defined zone of inhibition obtained, it is apparent that the functionalized samples are bacteriostatic and not bactericidal. In addition, the results also showed that the antibacterial effect of CH/ALG occurred without migration of the active agents. As CH/ALG are in a solid form, only microorganisms in direct contact with the active sites of CH/ALG are inhibited because CH/ALG layers are incapable of diffusing through the adjacent agar medium, as discussed elsewhere.^[29] As a result, and as described previously, the possible mechanisms for antibacterial activity are: (1) the CH/ALG on the surface of the cell can form a polymer membrane, which prevents nutrients from entering the cell. (2) CH/ALG entered the cell through pervasion, since CH could absorb the electronegative substance in the cell and flocculate them; it disturbs the physiological activities of the bacteria.^[30]

Assessment of antibacterial activity – JIS L 1902:2002 - absorption method

The antibacterial effect of functionalized samples was tested according to the Japanese Industrial Standard JIS L Standard 1902:2002. This method is based on the quantitative determination of the potential effect and activity of functionalized samples, by the direct contact with a suspension of bacterial cells. The results of the bioactivity investigations are presented in Table 1.

All samples (CT, CT5, CT6, CT9 and CT10) showed a bacteriostatic activity and no bactericidal activity found against *S. aureus* and *K. pneumoniae*, being in accordance with the results for antibacterial activity by the halo method. CT5 is the sample that has the highest value of bacteriostatic activity (1.9 for *S. aureus* and 1.5 for *K. pneumoniae*), followed by the sample CT9 (1.8 for *S. aureus* and 1.3 for *K. pneumoniae*). These samples have in common the last layer is composed for CH, suggesting more free amino groups from CT5 and CT9 and lower free amino groups from the CT6 and CT10, which are bonded with carboxylic groups of the ALG thus, reducing the activity. The experimental results provide tangible evidence in support of the hypothesis that the amino group on CH is a source of bacteriostatic activity.

Table 1. Antibacterial activity (bacteriostatic and bactericidal activity)

Sample	<i>S. aureus</i>		<i>K. pneumoniae</i>	
	$M_b - M_c$	$M_a - M_c$	$M_b - M_c$	$M_a - M_c$
CO	0	-1.5	0	-1.8
CT	0.3	-1.1	0.2	-1.0
CT5	1.9	-0.5	1.5	-0.4
CT6	1.3	-1.0	1.1	-0.9
CT9	1.8	-0.6	1.3	-0.6
CT10	1.4	-0.7	1.1	-0.8

A-number of inoculated bacteria. B-number of bacteria on the standard sample contacted for 18 h. C-number of bacteria on the functionalized sample after incubation for 18 h. $M_a = \log A$, $M_b = \log B$, $M_c = \log C$. Bacteriostatic activity level, $M_b - M_c$; bactericidal activity level, $M_a - M_c$.

In literature, normally the CH exhibits a stronger bioactivity effect upon Gram-positive than Gram-negative bacteria. This observation may be explained by the higher deacetylation degree of CH used in this work (80%), which implies a higher number of side amine groups available for reaction.^[4,27] On the other hand, the bioactivity effect also depends on the molecular weight; the inhibitory effect decrease slightly as the molecular weight increase. In this study, it was used CH of low molecular weight (420 kDa). CH is a cation that attracts the negative charges of the cell walls of bacteria, as claimed by several authors, being the cause for CH antibacterial action.^[31–34] The surface of *S. aureus* includes the negatively charged teichoic acid within a thick peptidoglycan layer that lacks an outer membrane.^[35] This should render the bacterial cell more attractive to and easier to be damaged through electrostatic-mediated contact-inhibition mechanisms when exposed to the positively charged CH layer (CT5 and CT9) than the negatively charged ALG layer (CT6 and CT10).

The lower bacteriostatic activity of the samples against *K. pneumoniae* than that against *S. aureus* may be explained as a result of the different bacterial membrane structures. In contrast to that of *S. aureus*, *K. pneumoniae* has a double protective layer, the outer lipopolysaccharide layer embedded with a number of small channels of porins and the inner peptidoglycan layer. The fact that it does not contain the negatively charged (teichoic acid) entities that can interact with positively charged molecules would potentially make it less sensitive to electrostatic binding with positively charged molecules like CH.^[35]

The work on the effect of CH in strains of *K. pneumoniae* and *S. aureus* found that CH promotes aggregation of bacterial cells and disorganization of bacterial cell wall and cytoplasmic membrane, which leads to the release of bacterial contents into the environment. These structural changes result in bacterial death.^[36] These results suggest that the apparent difference action upon Gram-positive and Gram-negative microorganisms probably results from the intrinsic difference in their cell wall structure.

All functionalized samples (CT5, CT6, CT9 and CT10) showed antibacterial activity against *S. aureus* and *K. pneumoniae* in solution. Figure 5 shows the growth inhibition (cell reduction) of the *S. aureus* and *K. pneumoniae* by the antibacterial activity of the functionalized samples with CH/ALG.

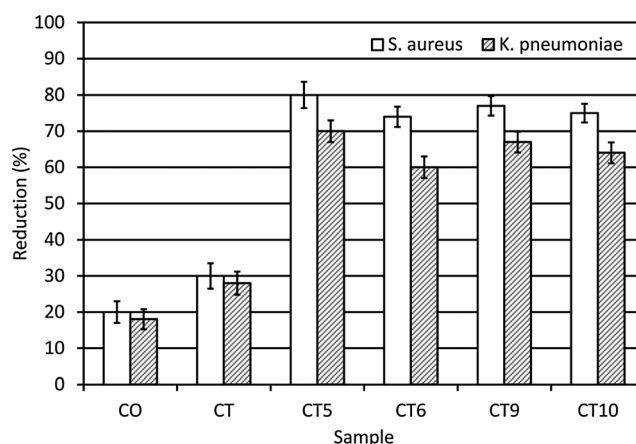


Figure 5. Reduction rate (%) of *Staphylococcus aureus* and *Klebsiella pneumoniae* in functionalized samples.

Analyzing the results, (Fig. 5) there is a reduction of 65–80% in bacterial growth on CT5 and CT9 samples, respectively, and a reduction of 60–75% in bacterial growth is achieved by CT6 and CT10 samples. An interesting observation is that all functionalized samples exhibit a high reduction of bacterial growth in solution although without a clear zone of inhibition assessed by the halo method. This may be because the concentration of CH and ALG on cotton samples is not sufficient enough for bactericidal activity as described elsewhere.^[37] Singh and co-workers found that antimicrobial efficacy of a compound will vary when it is present in solution and when it is held intimately by a textile substrate.^[37]

All of these results suggest that the functionalized samples with five numbers of layers (CH/ALG/CH/ALG/CH) are more active against *S. aureus* and *K. pneumoniae* microorganisms.

Structurally, the cationic nature of CH is expected to interact strongly by ionic bonds with the anionic ALG, and the combination of ALG with CH has become quite commonplace for the development of potential wound healing materials as they showed no toxic effects to mammalian cells.^[38] A recent study has shown that ALG/CH-based wound dressing films accelerated burn healing by modulating the epithelization, blood vessels formation and collagenization process.^[39]

CONCLUSION

The results obtained confirm the possibility of using the LbL method for modification of the surfaces of cotton fabric in order to impart antibacterial properties to them. The most important potential application of these materials could be their use as external wound dressings. Their advantage toward the existing materials is that they are fully biocompatible and inexpensive.

Many applications are proposed based on this CH/ALG, which is the most investigated polyelectrolyte complex, especially for biomedical applications. Note that this technique (LbL) in textile is entirely new and is a simple and effective method with strong possibility of industrial application.

This new coating for cellulosic fibers is a new strategy and may open new avenues for the development of antimicrobial polymers with potential application in health-care field.

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